

A Novel Reformulation of the Pseudo2D Battery Model Coupling Large Deformations at Particle and Electrode Levels

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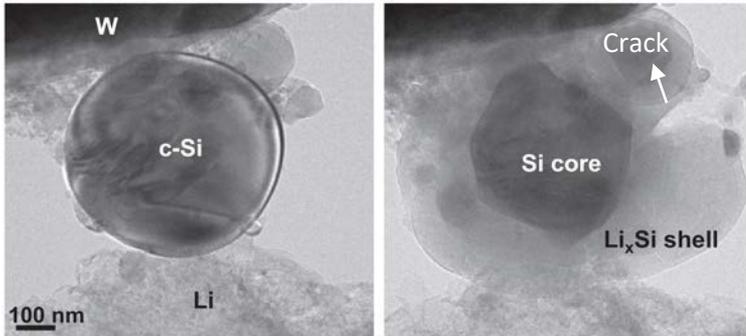
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Introduction

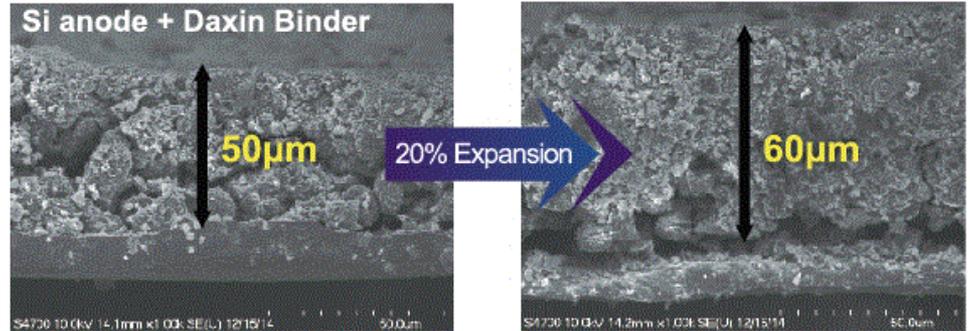
- ❑ Si anode has high energy density but suffer from large deformation

Huang et al. Acta Materialia (2013)



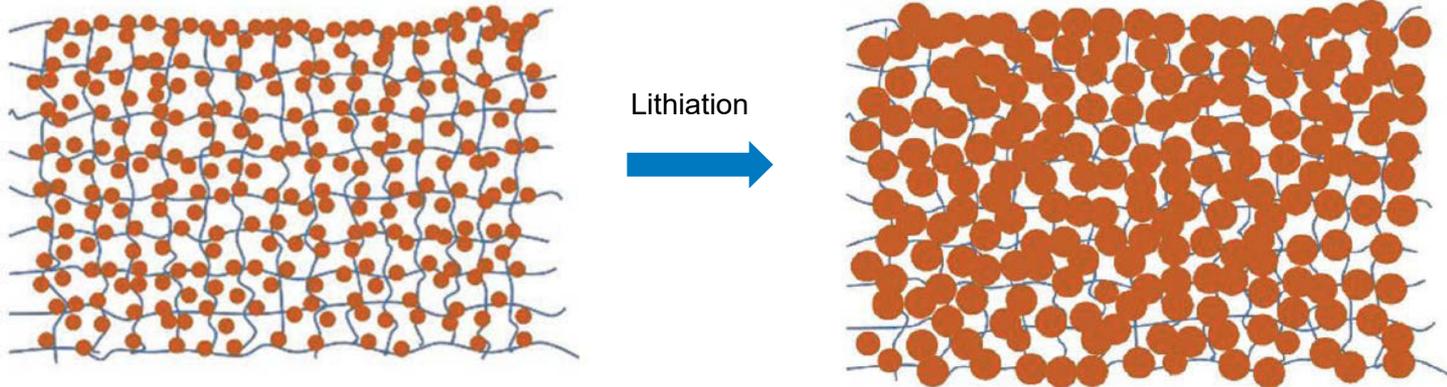
400% volume expansion of Si

Daxin (<http://www.daxinmat.com>)



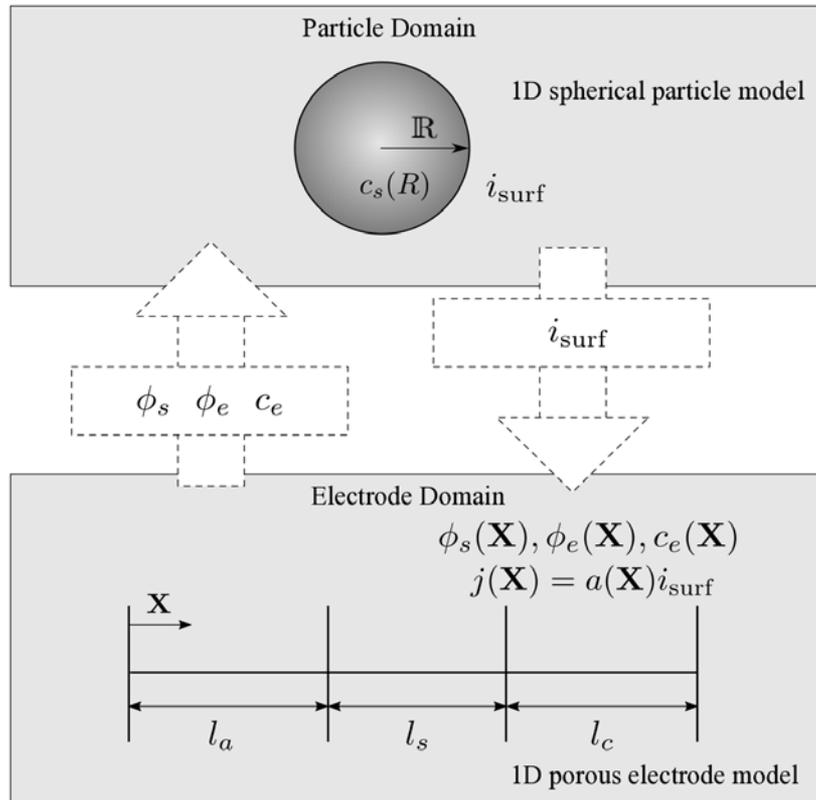
Electrode/Cell deformation

Wang et al. Advanced Energy Materials (2018)

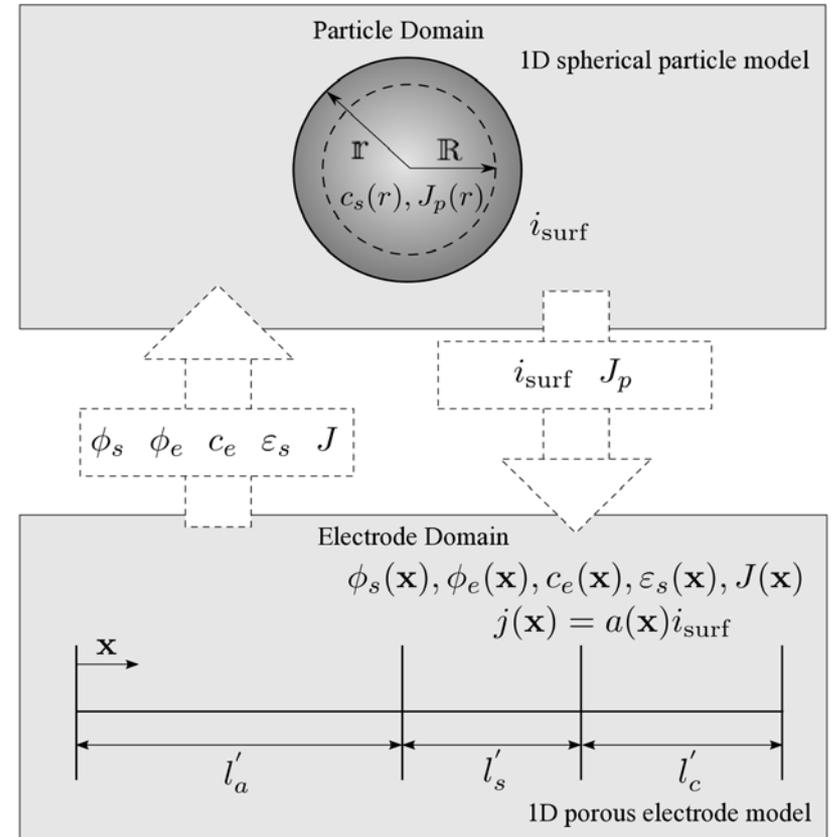


- ❖ Active material (AM) expansion causes electrode deformation and porosity reduction
- ❖ A model coupling multi-scale deformations required for better cell design

Introduction



P2D Newman model



P2D model coupling large deformations

- ❖ **Goal:** consistently incorporate deformations based on the P2D framework
- ❖ **Challenge:** infinitesimal deformation assumption inapplicable

Formulation: large deformation in electrode domain

□ Finite strain theory

- Deformation composed of elastic and inelastic deformations

$$\mathbf{F} = \mathbf{I} + \nabla \mathbf{u} \quad \mathbf{F} = \mathbf{F}_e \mathbf{F}_c \quad \mathbf{u} : \text{displacement vector}$$

multiplicative decomposition

\mathbf{F} : deformation gradient tensor

- Isotropic inelastic deformation due to Li insertion/extraction

$$\mathbf{F}_c = \left(1 + \frac{\Omega_e}{3} \Delta C_{s,\text{avg}} \right) \mathbf{I} \quad \epsilon_e = \frac{1}{2} (\mathbf{F}_e^T \mathbf{F}_e - \mathbf{I})$$

Ω_e : partial molar volume of Li in electrode
 ϵ_e : elastic strain tensor

- Displacement can be solved by

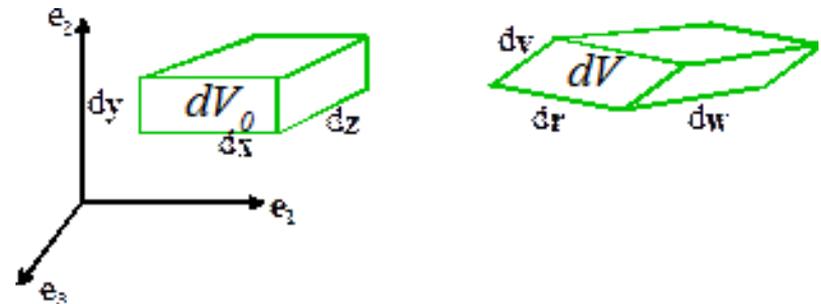
$$\mathbf{S} = J_c \mathbf{F}_c^{-T} (\mathbf{C} : \epsilon_e) \mathbf{F}_c^{-1} \quad \nabla \cdot (\mathbf{F}\mathbf{S})^T = 0$$

\mathbf{S} : Secondary PK stress
 σ : Cauchy stress
 \mathbf{C} : stiffness tensor

$$\sigma = J^{-1} \mathbf{F}\mathbf{S}\mathbf{F}^T$$

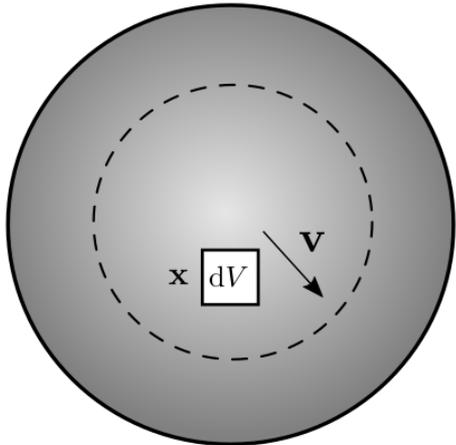
- The **Jacobian** of the deformation gradient – change of volume

$$J = \det(\mathbf{F}) = \frac{dV}{dV_0}$$



Formulation: conservation law in reference frame

□ Eulerian conservation law

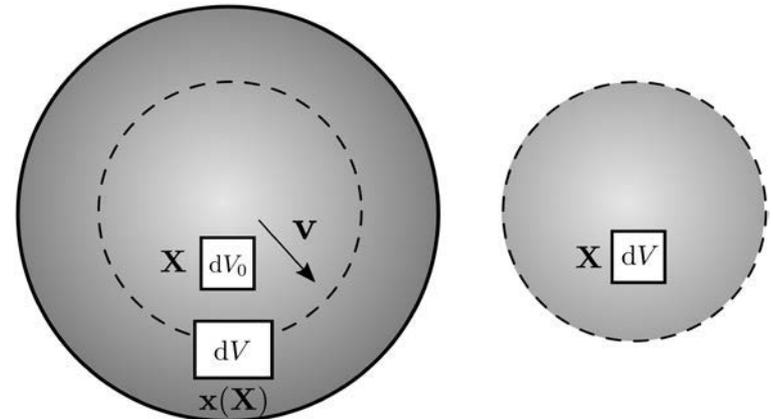


$$\frac{\partial c(\mathbf{x})}{\partial t} = -\nabla_x \cdot \mathbf{N}(\mathbf{x}) + R(\mathbf{x})$$

$$\mathbf{N}(\mathbf{x}) = -D\nabla_x c(\mathbf{x}) + c(\mathbf{x})\mathbf{v}(\mathbf{x})$$

- Volume element (fixed in space)
- Need to include a convection term
- Need to explicitly keep track of the deformation

□ Lagrangian conservation law



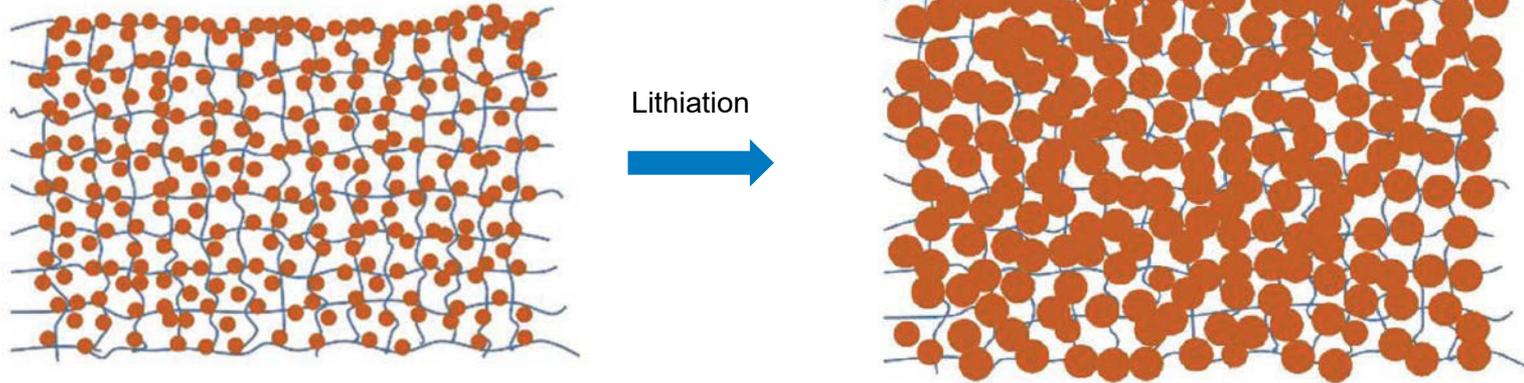
$$\frac{\partial}{\partial t} \left[c(\mathbf{X})J(\mathbf{X}) \right] = -\nabla_X \cdot \mathbf{N}(\mathbf{X}) + R(\mathbf{X})J(\mathbf{X})$$

$$\mathbf{N}(\mathbf{X}) = -J\mathbf{F}^{-1}D\mathbf{F}^{-T}\nabla_X c(\mathbf{X}) = -D_X\nabla_X c(\mathbf{X})$$

- Approximate field distributions in the undeformed geometry
- Material volume: $dV_0(\mathbf{X}) \rightarrow dV(\mathbf{x}(\mathbf{X}))$
- Effect of deformation on conservation is embodied in deformation gradient tensor \mathbf{F}

Formulation: porosity variation

Wang et al. Advanced Energy Materials (2018)



- AM expansion causes porosity reduction and electrode deformation

$$\left[\frac{\partial \varepsilon_s}{\partial t} \right] + \left[\nabla \cdot (\varepsilon_s \mathbf{v}) \right] = \left[-\frac{s\Omega_e}{nF} j \right]$$

Variation rate of solid volume fraction Electrode deformation rate Increase rate of AM volume

ε_s : volume fraction of solid phase

Ω_e : partial molar volume of Li in electrode

\mathbf{v} : local electrode velocity vector

$j = a(\mathbf{x})i_{\text{surf}}$: volumetric current source

- Ratio of porosity reduction and electrode deformation depends on fixture condition
- Material frame reformulation

$$\frac{\partial(\varepsilon_s J)}{\partial t} = -\frac{s\Omega_e}{nF} j J$$

Formulation: P2D with large deformation

□ Particle deformation

- Particle size change from \mathbb{R} to $\mathbb{r}(\mathbf{X})$ after lithiation/delithiation
- Within each particle, the deformation is characterized by the particle deformation gradient tensor $\mathbf{F}_p(\mathbf{R})$

$$\mathbf{F}_p = \begin{bmatrix} \frac{\partial r}{\partial R} & 0 & 0 \\ 0 & \frac{r}{R} & 0 \\ 0 & 0 & \frac{r}{R} \end{bmatrix}$$

- In the current model, we assumed that deformation within particle is uniform

$$\frac{\partial r}{\partial R} = \frac{r}{R} = \lambda \quad \longrightarrow \quad J_p = \frac{V_p}{V_{p,0}} = \det(\mathbf{F}_p) = \lambda^3$$

- Alternatively the particle deformation can be expressed in terms of electrode-level variables

$$J_p = \frac{dV_s}{dV_{s,0}} = \frac{\varepsilon_s}{\varepsilon_{s,0}} J$$

- Particle stretch can be expressed as

$$\lambda = \frac{r}{R} = \left(\frac{\varepsilon_s}{\varepsilon_{s,0}} J \right)^{1/3}$$

AM expansion affects solid diffusion distance

Formulation: P2D with large deformation

- ❑ Solid diffusion in particle

$$\frac{\partial}{\partial t}(J_p c_s) = -\frac{1}{R^2} \nabla_L (R^2 \mathbf{J}_L)$$

- ❑ Charge conservation in electrolyte

$$\nabla_L \cdot \mathbf{i}_l = jJ$$

- ❑ Charge conservation in electrodes

$$\nabla_L \cdot \mathbf{i}_s = -jJ$$

- ❑ Mass conservation in electrolyte

$$(1 - \varepsilon_s) J \frac{\partial c_e}{\partial t} = \nabla_L \cdot [D_t^L \nabla_L c_e - \frac{\mathbf{i}_e t_+}{F}] + \frac{j}{nF} J.$$

electrolyte modeled as incompressible fluid

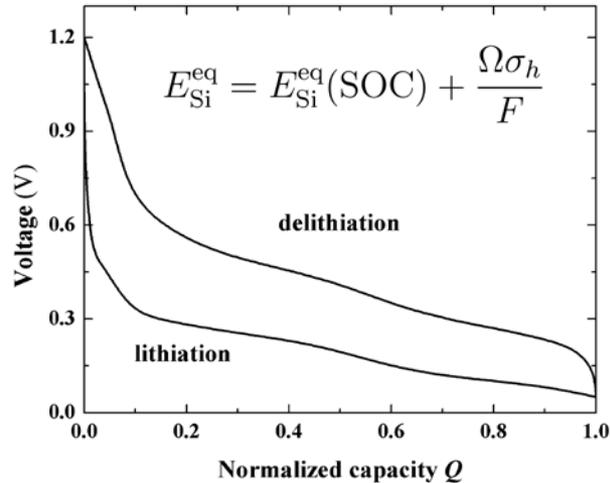
The new model:

- Approximates two additional fields (electrode displacement, porosity)
- Only requires minor modifications of the existing P2D governing equations

Additional multiphysics coupling and assumptions

□ Stress-dependent OCP

Lu et al. Physical Chemistry Chemical Physics (2016)



Voltage hysteresis of Li_xSi system due to the effect of stress

□ Porosity-dependent mechanical properties

Kovacik et al. Journal of materials science letters (1999)

$$E = E_s \left(1 - \frac{\varepsilon_e}{\varepsilon_0}\right)^n$$

$$\nu = \nu_s + \frac{\varepsilon_e}{\varepsilon_1} (\nu_0 - \nu_s)$$

□ Specific surface area

$$a = \frac{3\varepsilon_s}{r(x)} = \frac{3\varepsilon_s}{\mathbb{R}} J_p^{-\frac{1}{3}}(x)$$

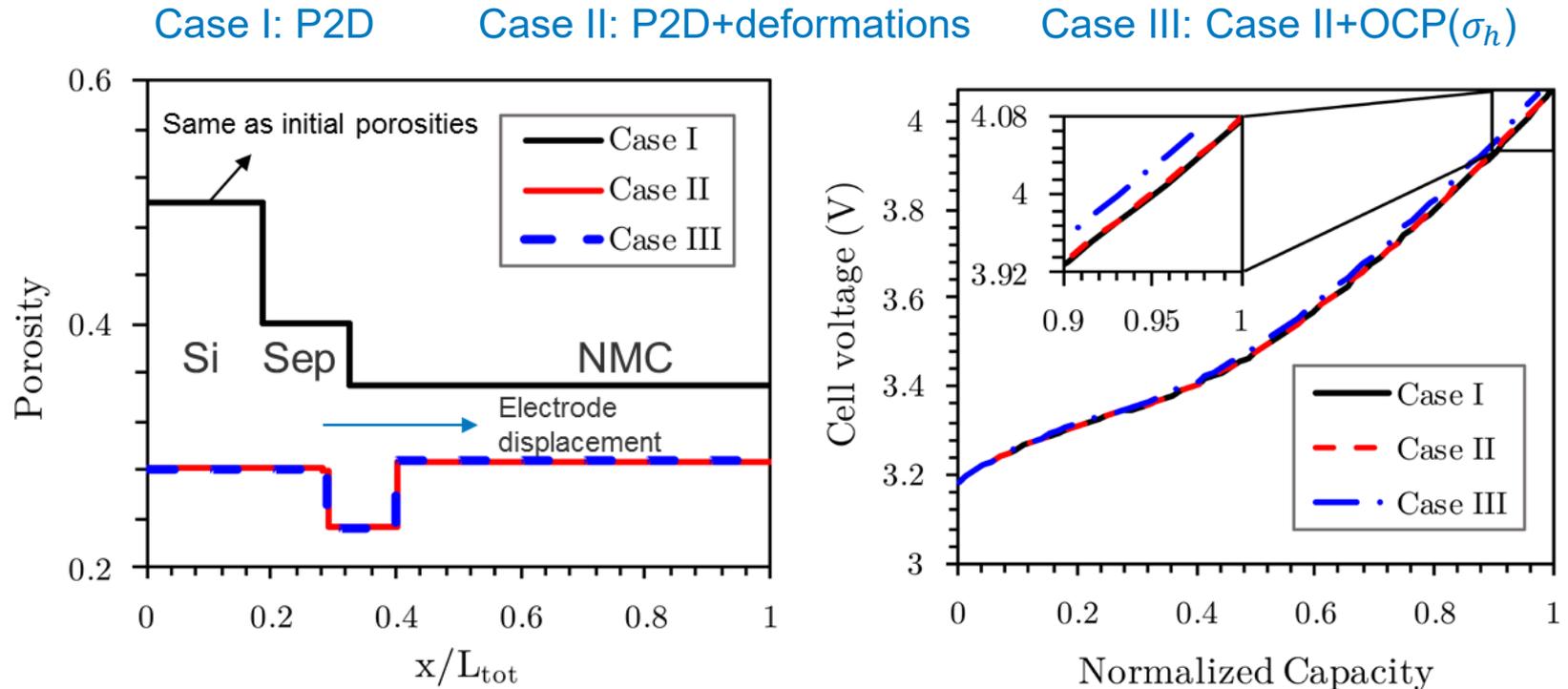
Couples particle deformation and porosity reduction

□ Assumptions

- All deformations are elastic and nondestructive
- Uniform and isotropic deformation within each particle
- Negligible in-plane electrode deformation (electrode is well adhered to strong metal foil cc)
- Electrolyte move out/into a material volume only in the out-of-plane direction
- Electrode is composed of only active material and electrolyte

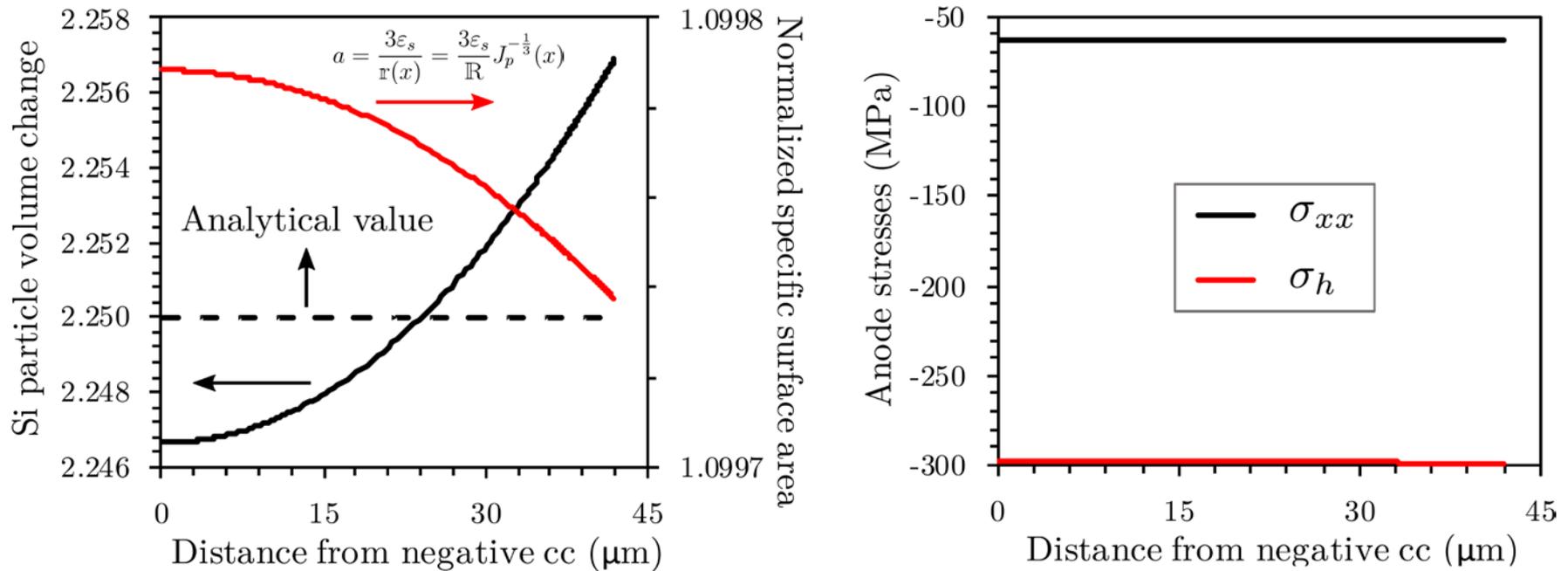
Low rate performance (0.02C)

- ❖ Si anode/NMC532 cathode; ANL Gen2 electrolyte; 5 mAh/cm² ($L_{cell} = 143.3 \mu\text{m}$)
- ❖ 0.02C charge to 4.08 V; both ends of the cell are fixed



- Thickness changes: anode (35.6%↑), separator (20.1%↓), cathode (9.3%↓)
- Porosity reductions: anode (43.8%↓), separator (41.8%↓), cathode (18%↓)
- Uniform porosity within each component
- Negligible impact on cell voltage and capacity

Low rate performance (0.02C, Case II)



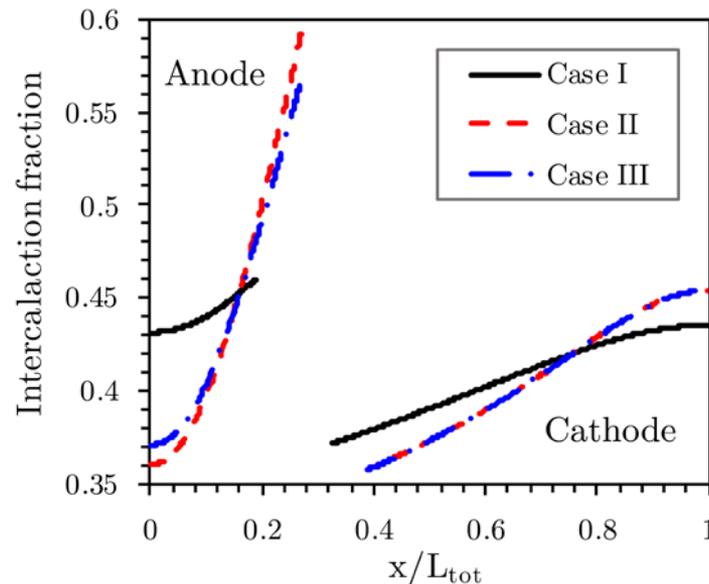
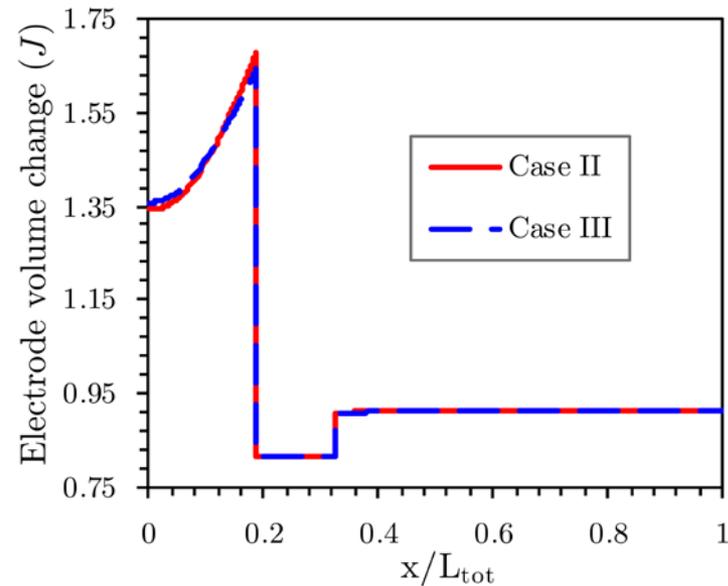
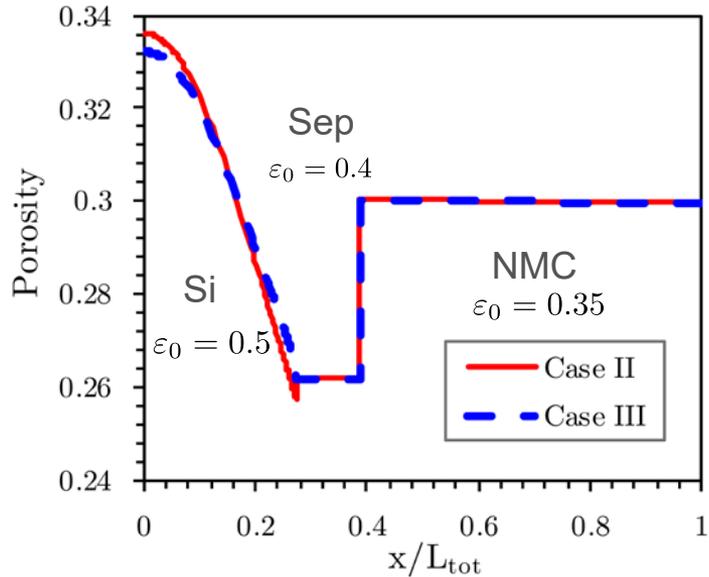
- Nonuniform particle expansion and specific surface area increase
- Magnitude of variation is small due to low charge rate
- Average particle expansion close to the analytical value
- Both σ_{xx} and σ_h in anode are uniform due to relatively uniform Li insertion rate distribution

High rate performance (1C)

Case I: P2D

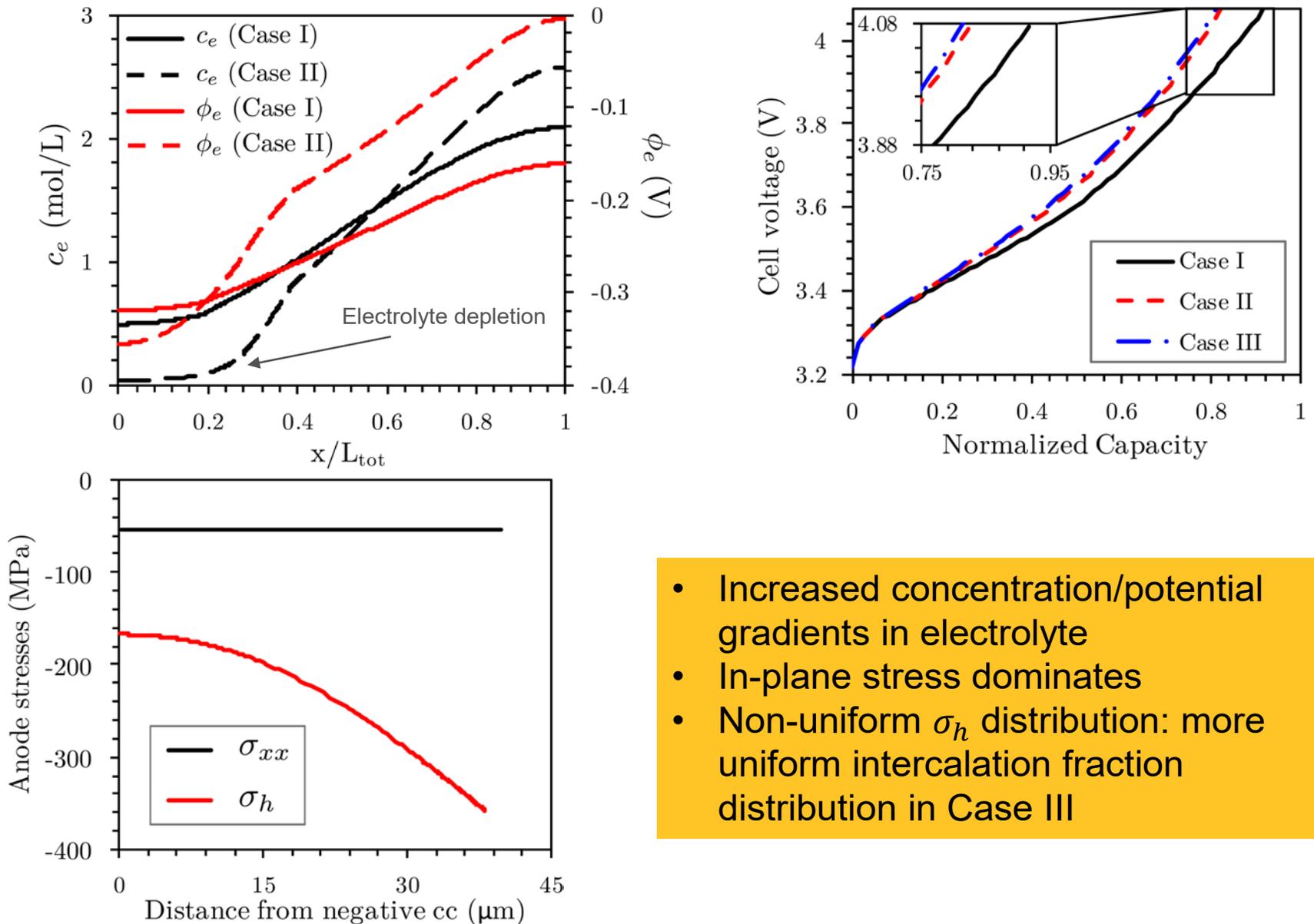
Case II: P2D+deformations

Case III: Case II+OCP(σ_h)



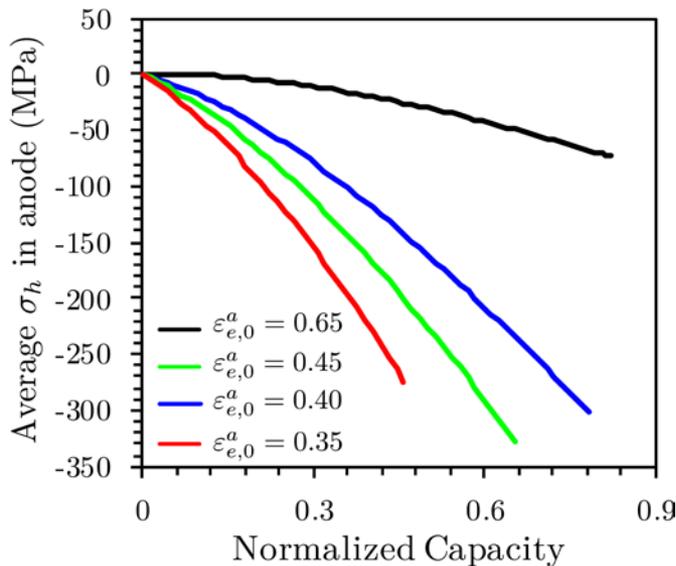
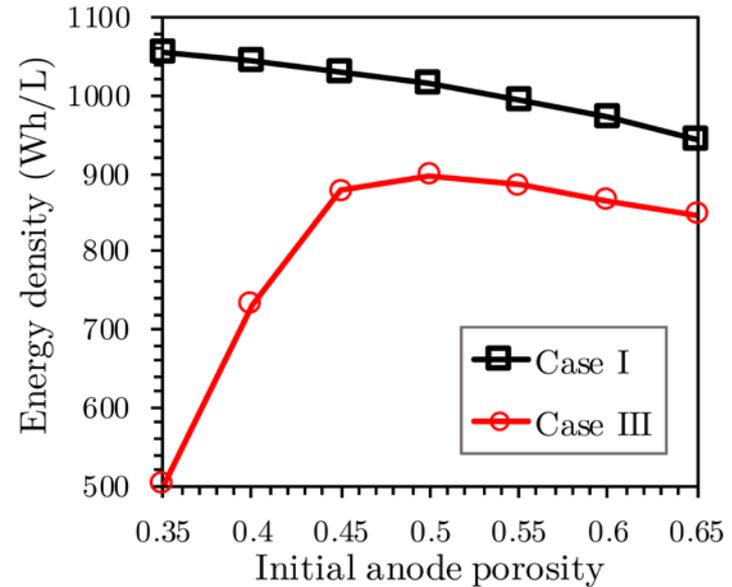
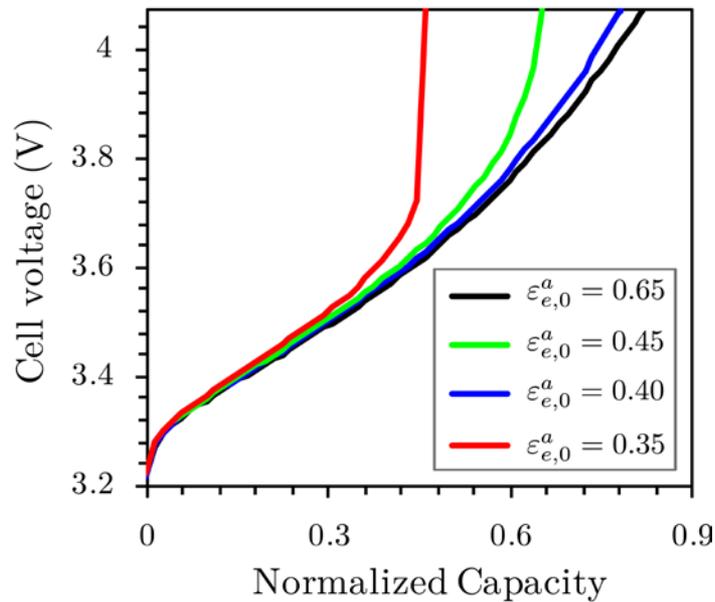
- Nonuniform porosity reduction and deformations due to faster lithiation rate near anode/separator interface
- Porosity reduction further decrease the utilization of active material near the anode/current collector interface
- Case III slightly improve the uniformity of field distributions in anode

High rate performance (1C)



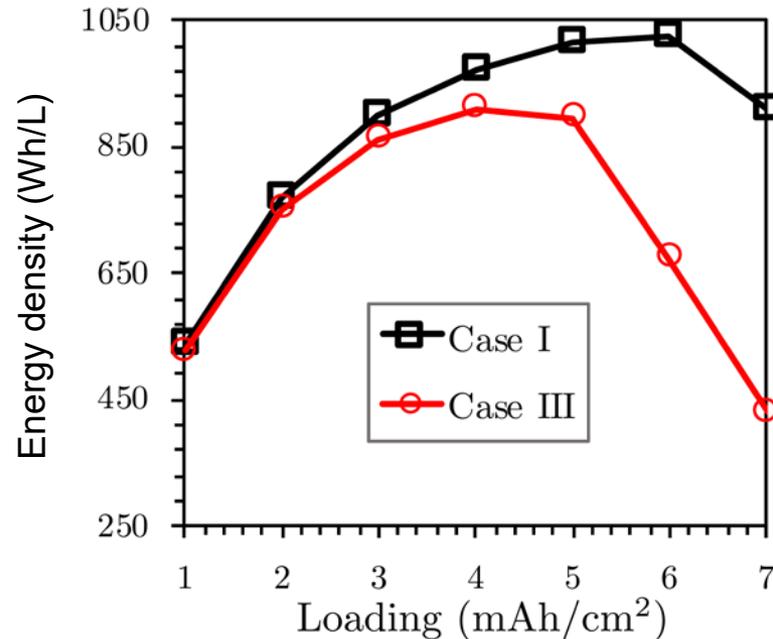
- Increased concentration/potential gradients in electrolyte
- In-plane stress dominates
- Non-uniform σ_h distribution: more uniform intercalation fraction distribution in Case III

Effect of anode porosity (1C, Case III)



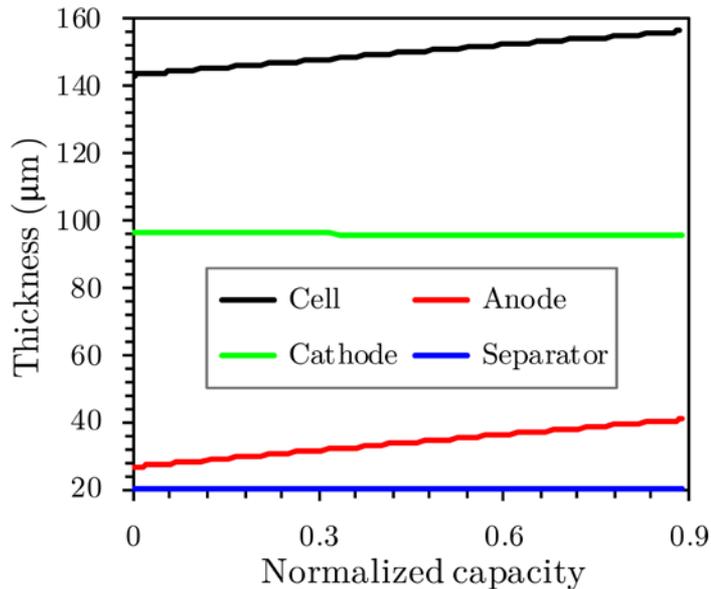
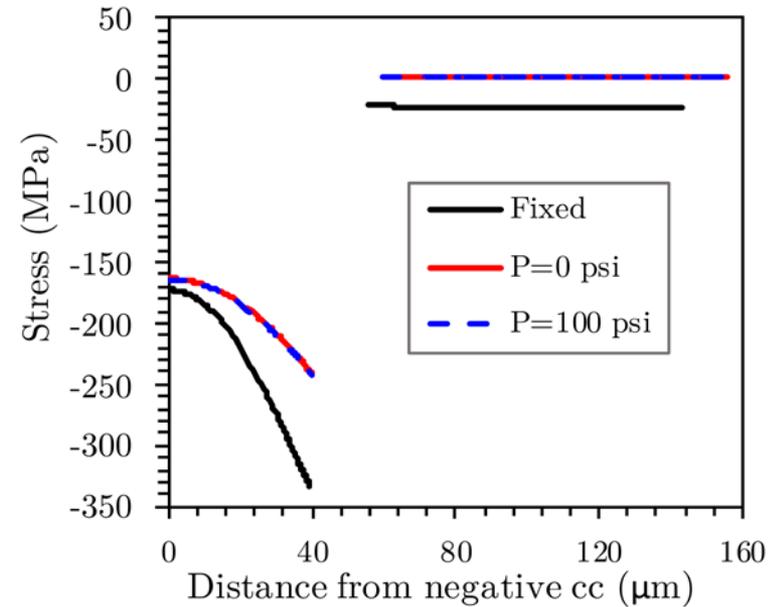
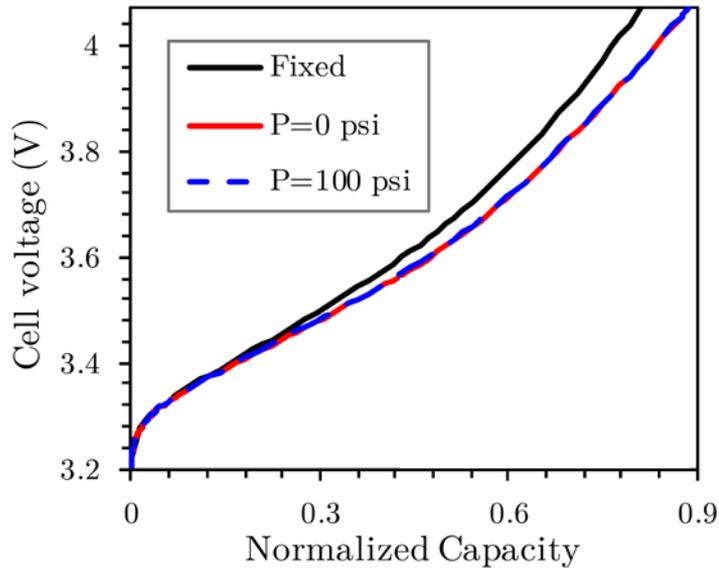
- Fixed loading (5 mAh/cm²); thicker electrode for lower initial porosity
- Optimal volumetric energy density (~900 Wh/L) obtained for $\epsilon_{a,0} = 0.5$
- Increasing the initial porosity helps to reduce σ_h in anode at the same state of charge

Effect of cell loading (1C, Case III)



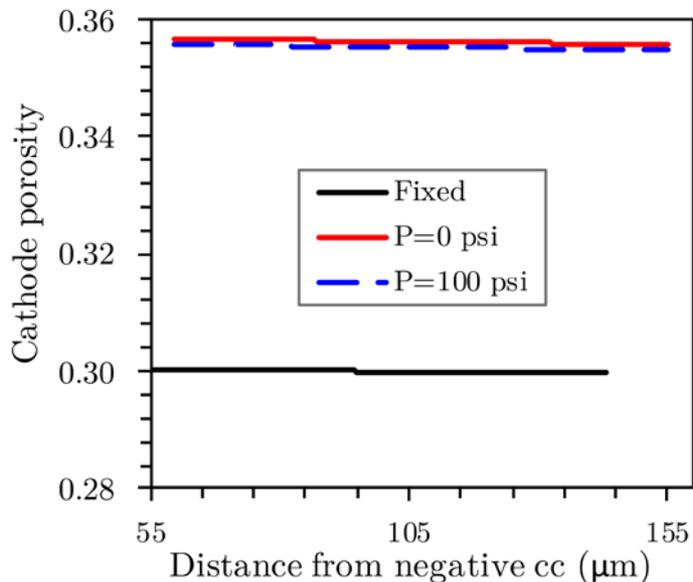
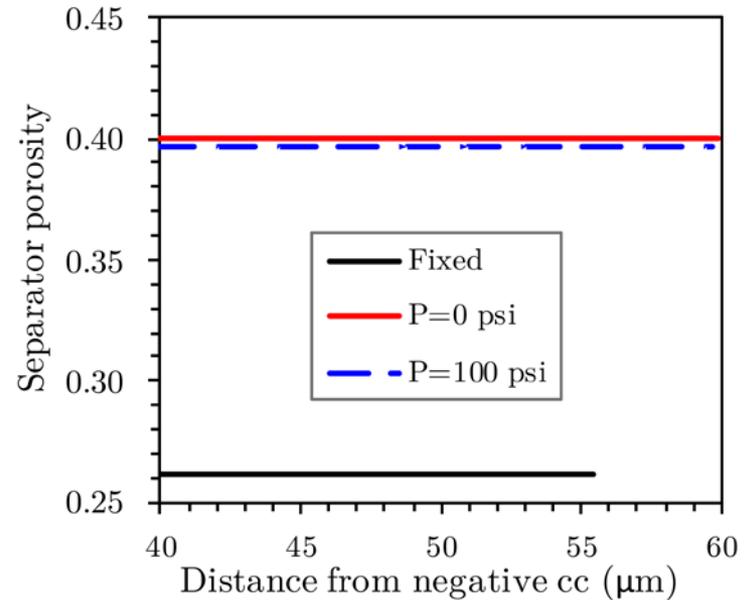
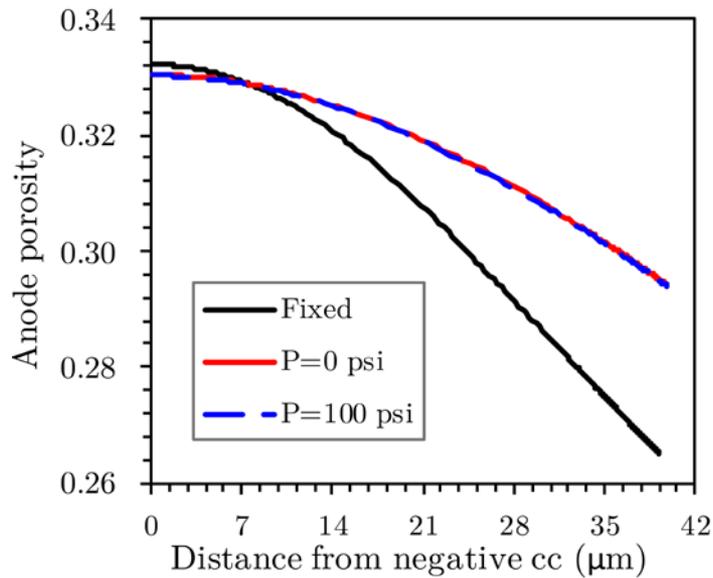
- Fixed initial anode porosity ($\varepsilon_{a,0} = 0.5$)
- Higher loading leads to thicker cell
- Classic P2D overpredicts cell energy density especially for higher loadings
- The predicted optimal loading is 4 mAh/cm²

Effect of cell fixture condition ($5 \text{ mAh/cm}^2, \epsilon_{a,0} = 0.5, 1C$)



- Higher cell capacity and lower electrode stress when the cell is free to expand
- Stress in cathode is slightly tensile for P=0 psi due to NMC contraction
- ~9.1% increase of cell thickness, mainly due to Si anode expansion

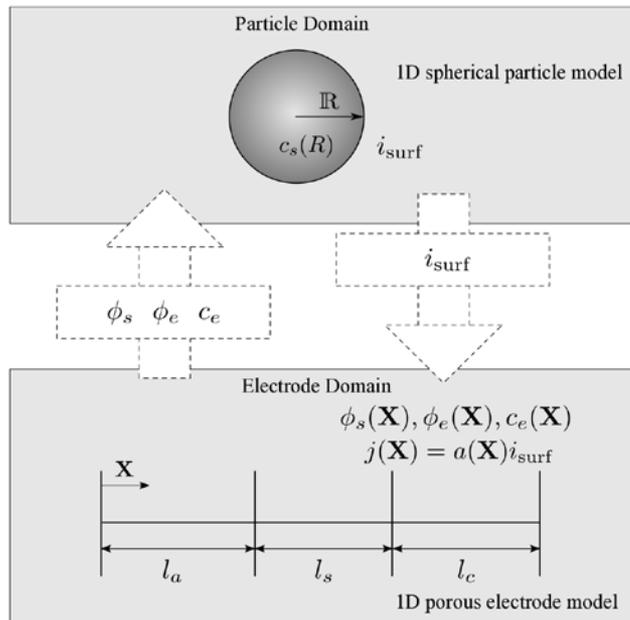
Effect of cell fixture condition ($5 \text{ mAh/cm}^2, \varepsilon_{a,0} = 0.5, 1\text{C}$)



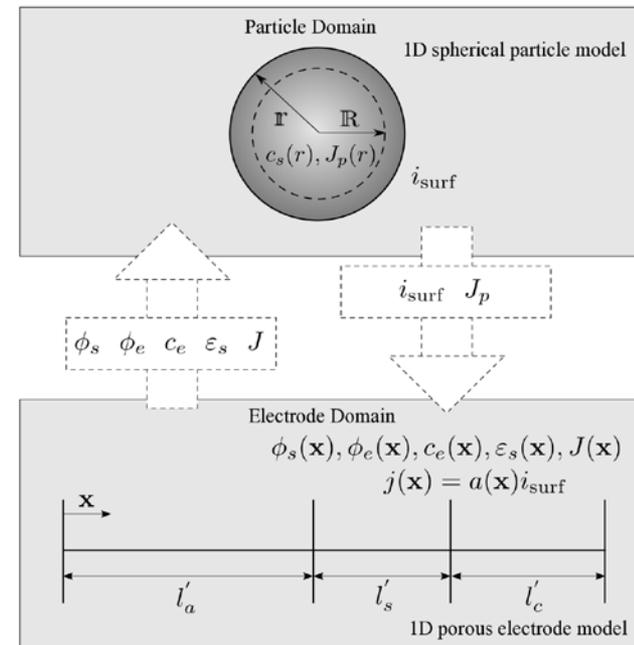
- Smaller porosity variation and thus more uniform Si utilization when $P=0$ psi
- Negligible porosity reduction in cathode and separator
- Separator is compressed more when both ends are fixed due to its lower Young's module compared to electrodes

Conclusion and future work

- The P2D model was reformulated to consistently couple particle and electrode deformations
- Deformations and porosity reduction significantly affects the accessible capacity of the cell
- The proposed model shows notable differences on predicting the optimal cell loading and electrode porosity compared with the P2D model
- The model is under further development to resolve particle-level stress and allow simulating performances of composite anode (Si/C)



P2D Newman model



P2D model coupling large deformations

Q&A

www.nrel.gov

NREL/PR-5400-73575

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Complementary materials

Parameter	Cathode	Separator	Anode
\mathbb{R} (μm)	1.8	N/A	0.1
D_s (m^2/s)	Appendix B	N/A	1e-16
κ_s (S/m)	100	N/A	100
i_0 (A/m^2)	Appendix B	N/A	1
Ω (m^3/mol)	7.8e-7 [22]	N/A	9.0e-6 [23]
$C_{s,\text{max}}$ (kmol/m^3)	49.6	N/A	333.3
$\varepsilon_{e,0}$	0.35	0.4	0.5
L_0 (μm) @ 5 mAh/cm ² , N:P=1.2	96.4	20	26.9
Intercalation fraction	(0.3,0.9)	N/A	(0.1,0.6)
E_s (GPa)	2.5	1	5
ν	0.3	0.3	0.3
Bruggeman factor	2.2	2.5	2.2

Table 1: Values of the parameters used in the current model for all example problems unless stated otherwise.

Complementary materials

Variable	Governing equation
c_s	$\frac{\partial}{\partial t} \left[\frac{\varepsilon_s}{\varepsilon_{s,0}} \left(1 + \frac{\partial u}{\partial X} \right) c_s \right] = \frac{1}{R^2} \frac{\partial}{\partial R} \left[R^2 D_E \left[\frac{\varepsilon_s}{\varepsilon_{s,0}} \left(1 + \frac{\partial u}{\partial X} \right) \right]^{1/3} \frac{\partial c_s}{\partial R} \right]$
ε_s	$\frac{\partial}{\partial t} \left[\left(1 + \frac{\partial u}{\partial X} \right) \varepsilon_s \right] = -\frac{s\Omega}{nF} \left(1 + \frac{\partial u}{\partial X} \right) i_E a$
ϕ_s	$\begin{aligned} \frac{\partial i_s}{\partial X} &= -\left(1 + \frac{\partial u}{\partial X} \right) i_E a \\ i_s &= -\kappa_{s,\text{eff}} \nabla \phi_s, \quad \kappa_{s,\text{eff}} = \kappa_s \varepsilon_s^b / \left(1 + \frac{\partial u}{\partial X} \right) \end{aligned}$
ϕ_e	$\begin{aligned} \frac{\partial i_e}{\partial X} &= \left(1 + \frac{\partial u}{\partial X} \right) i_E a \\ i_e &= -\kappa_{e,\text{eff}} \nabla \phi_e + \left(\frac{2\kappa_{e,\text{eff}} RT}{F} \right) \left(1 + \frac{\partial \ln f_{\pm}}{\partial \ln c_e} \right) (1 - t_+) \nabla \ln c_e \\ \kappa_{e,\text{eff}} &= \kappa_e \varepsilon_e^b / \left(1 + \frac{\partial u}{\partial X} \right) \end{aligned}$
c_e	$\begin{aligned} \varepsilon_e \left(1 + \frac{\partial u}{\partial X} \right) \frac{\partial c_e}{\partial t} &= \frac{\partial}{\partial X} \left(D_{e,\text{eff}} \frac{\partial c_e}{\partial X} - \frac{i_e t_+}{F} \right) + \frac{s}{nF} \left(1 + \frac{\partial u}{\partial X} \right) i_E a \\ D_{e,\text{eff}} &= D_e \varepsilon_e^b / \left(1 + \frac{\partial u}{\partial X} \right) \end{aligned}$
u	$\begin{aligned} \nabla(\mathbf{FS})_{XX} &= 0 \\ (\mathbf{FS})_{XX} &= \left(1 + \frac{\partial u}{\partial X} \right) \left(1 + \frac{\Omega \Delta C_{s,\text{avg}}}{3} \right) \frac{E(1-\nu)}{2(1+\nu)(1-2\nu)} \left[\left(\frac{1 + \frac{\partial u}{\partial X}}{1 + \frac{\Omega \Delta C_{s,\text{avg}}}{3}} \right)^2 + \right. \\ &\quad \left. \frac{2\nu}{(1-\nu)(1 + \frac{\Omega \Delta C_{s,\text{avg}}}{3})^2} - \frac{1+\nu}{1-\nu} \right] \end{aligned}$

Table 3: Explicit forms of the governing equations. Derivatives are defined in the reference configuration.

High rate performance (1C)

