



Co-Optimization of
Fuels & Engines

Co-Optimization of Fuels & Engines (Co-Optima) Initiative

John Farrell, National Renewable Energy Lab

SAE 13th International Conference on
Engines & Vehicles - Capri, Italy
September 13, 2017

NREL/PR-5400-70200



better fuels | better vehicles | sooner

U.S. DEPARTMENT OF
ENERGY

Energy Efficiency &
Renewable Energy

Acknowledgments



DOE Sponsors:

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Kevin Stork, Gurpreet Singh, Leo Breton, and
Mike Weismiller (VTO)



Co-Optima Technical Team Leads:

Dan Gaspar (PNNL), Paul Miles (SNL), Jim Szybist (ORNL),
Jennifer Dunn (ANL), Matt McNenly (LLNL), Doug Longman (ANL)

Other Co-Optima Leadership Team Members:

John Holladay (PNNL), Robert Wagner (ORNL), Chris Moen (SNL)

Goal: better
fuels and
better vehicles
sooner



Fuel and Engine Co-Optimization

- What fuel properties maximize engine performance?
- How do engine parameters affect efficiency?
- What fuel and engine combinations are sustainable, affordable, and scalable?



Central Engine Hypothesis

There are engine architectures and strategies that provide higher thermodynamic efficiencies than are available from modern internal combustion engines; new fuels are required to maximize efficiency and operability across a wide speed / load range



Central Fuel Hypothesis

If we identify target values for the critical fuel properties that maximize efficiency and emissions performance for a given engine architecture, then fuels that have properties with those values (regardless of chemical composition) will provide comparable performance



Two Parallel R&D Projects



Light-Duty



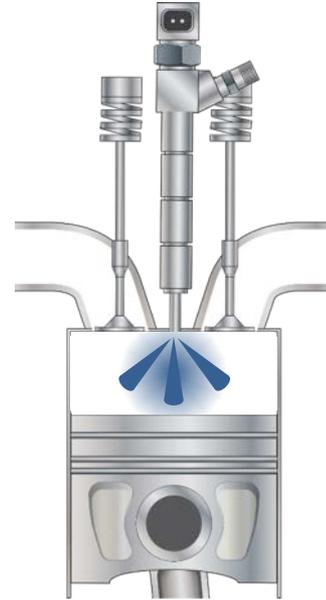
Boosted SI

Near-term



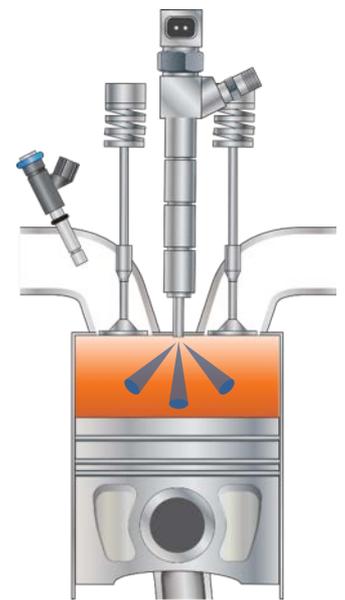
Multi-mode SI/ACI

Mid-term



Mixing Controlled

Near-term



**Kinetically
Controlled**
Longer-term

High-level goals and outcomes



Light-duty

Up to 15% fuel economy (FE) improvement*
boosted SI and multi-mode SI/ACI

Heavy-duty

Up to 4% FE improvement (worth \$5B/year)*
Potential lower cost path to meeting next tier
of criteria emissions regulations

Fuels

Diversifying resource base

Providing economic options to fuel providers
to accommodate changing global fuel demands

Increasing supply of domestically sourced
fuel by up to 25 billion gallons/year

Cross-cutting goals

Stimulate domestic economy

Adding up to 500,000 new jobs

Providing clean-energy options

* Beyond projected results of current R&D efforts. The team is actively engaging with OEMs, fuel providers, and other key stakeholders to refine goals and approaches to measuring fuel economy improvements

Co-Optima Team



Nine national labs,
13 universities

> 100 researchers,
> 75 projects

External Advisory Board

77 stakeholder
organizations

Budget: FY16: \$26M

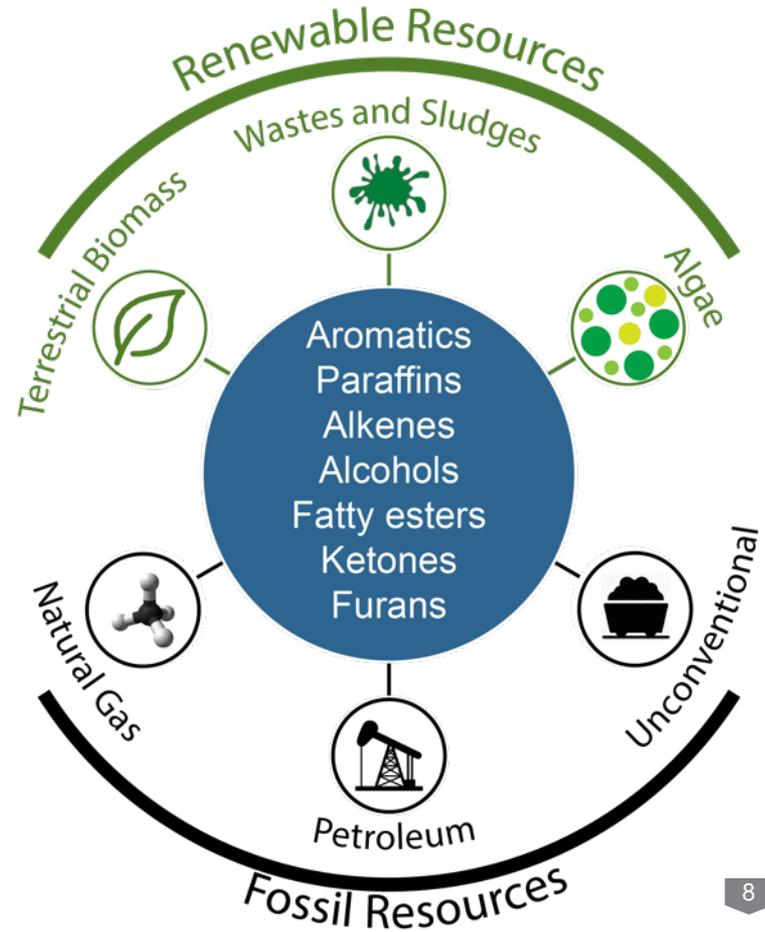
FY17: \$24.5

Approach



Objective: identify fuel properties that optimize engine performance, independent of composition,* allowing the market to define the best means to blend and provide these fuels

* We are not going to recommend that any specific blendstocks be included in future fuels



Systematic Blendstock Survey



Objective: identify a broad range of feasible blendstock options

Primary focus: identify blendstocks with desired properties that have a strong potential to be sourced from biomass

Potential Benefits of Biomass Sourced Fuel

Technical

Tailor fuel properties desired in the blendstock

Add value to refiners – blend up low quality (inexpensive) petroleum blendstocks

Help refiners balance global trends in transportation fuel use



Societal

Reliable domestic energy options that are affordable & efficient

Strengthens energy security by increasing supply, diversity, reliability

Retain \$260 billion in the U.S.

Add 1.1M direct jobs

Expand U.S. science/technology leadership



Environmental

Reduce emissions, including CO₂ emissions, by 450 million tons (7%) annually

Improved soil, water, and air quality



Introducing New Fuels and Engines Impacts a Large Body of Stakeholders



Energy Companies



Refiners



Biofuel Producers



Fuel Distribution



Government/
Regulatory Agencies



LD OEMs



HD OEMs



Retail



Consumer



Society

External Advisory Board



USCAR

David Brooks

American Petroleum Institute

Bill Cannella

Fuels Institute

John Eichberger

Truck & Engine Manufacturers Assn

Roger Gault

Advanced Biofuels Association

Michael McAdams

Flint Hills Resources

Chris Pritchard

EPA

Paul Machiele

CA Air Resources Board

James Guthrie

UL

Edgar Wolff-Klammer

University Experts

Ralph Cavalieri (WSU, emeritus)

David Foster (U. Wisconsin, emeritus)

Industry Expert

John Wall (Cummins, retired)

Co-Optima Scope



- Focusing only on liquid fuels
- Identify blendstocks to blend into petroleum base fuel
- Considering only non-food-based biofuel feedstocks
Assessing WTW emissions for biofuel options (GHG, water, etc)
- Considering hybridized/non-hybridized solutions
- Provide data, tools, and knowledge to stakeholders - objective is not to “pick winners”

Two Parallel R&D Projects



Light-Duty



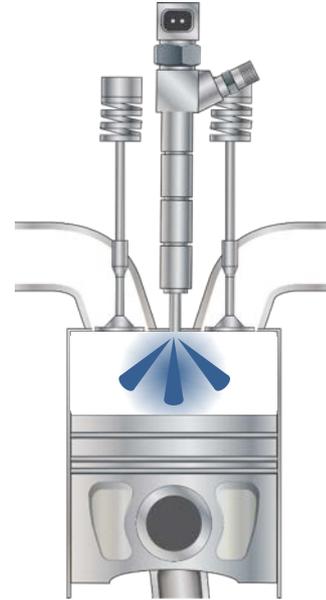
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Near-term



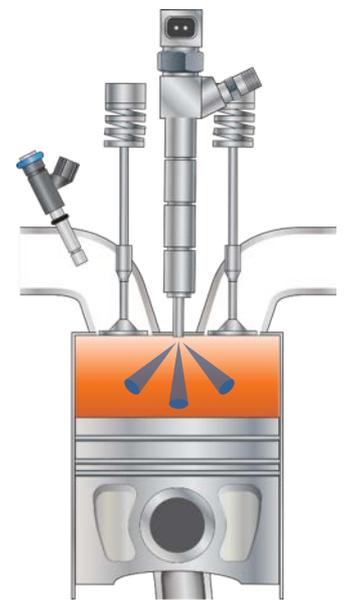
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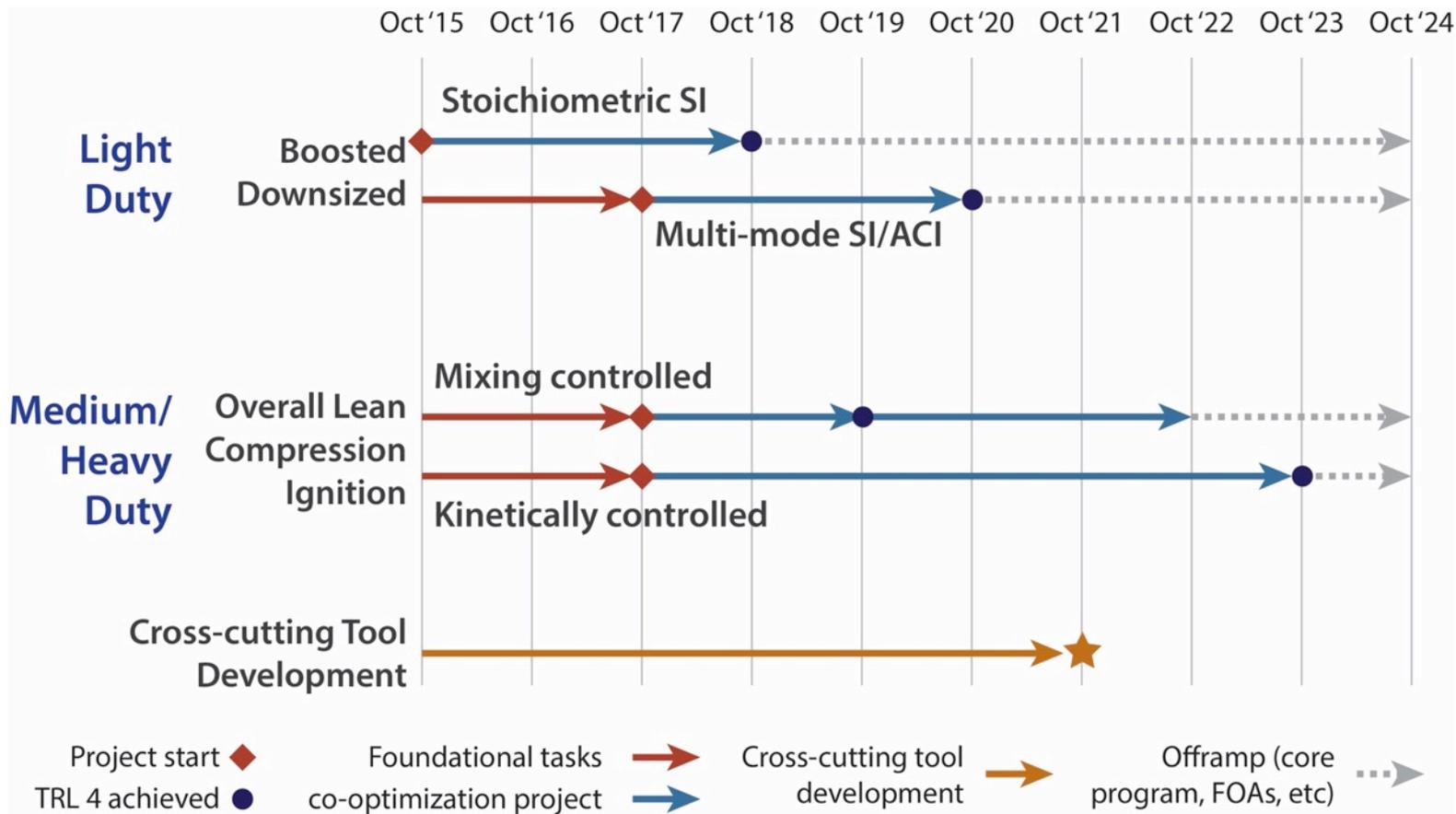
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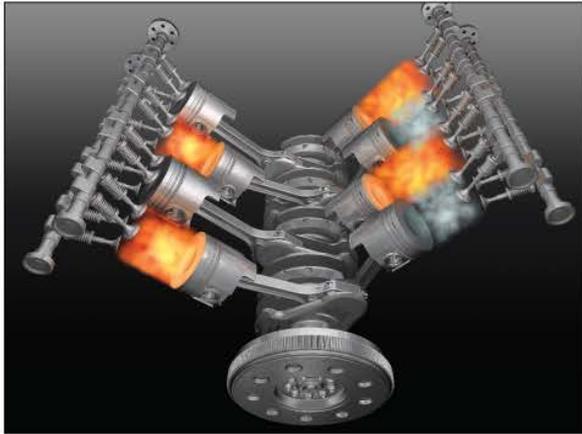
Timeline



Foundational Technical Questions



What fuels do
engines
really want?



What fuels
should we make?



What will work
in the real world?



Question 1: What fuels do engines really want?



Approach:

Conduct engine experiments and simulations that delineate fuel property impacts on engine performance



Theoretical foundation: “merit function”



Engine efficiency can be expressed as a product of various “efficiencies”:

$$\eta_{th} = \eta_{ideal} * \eta_{glh} * \eta_{comb} * \eta_{pump} * \eta_{ht} * \eta_{emiss} \dots$$

$$\eta_{ideal} = 1 - \frac{1}{CR^{\gamma-1}}$$

η_{glh} = combustion phasing (“degree of constant V combustion”)

η_{comb} = combustion efficiency

η_{pump} = pumping losses

η_{ht} = heat transfer losses

η_{emiss} = emission control losses

Theoretical foundation: “merit function”



Since we are interested in relative efficiency, we can differentiate to get:

$$\frac{d\eta_{th}}{\eta_{th}} = \frac{d\eta_{CR}}{\eta_{CR}} + \frac{d\eta_{\gamma}}{\eta_{\gamma}} + \frac{d\eta_{glh}}{\eta_{glh}} + \frac{d\eta_{comb}}{\eta_{comb}} + \frac{d\eta_{pump}}{\eta_{pump}} + \frac{d\eta_{ht}}{\eta_{ht}} + \frac{d\eta_{emiss}}{\eta_{emiss}} + \dots$$

The diagram shows the following connections from the terms in the equation:

- $\frac{d\eta_{CR}}{\eta_{CR}}$ (blue circle) points to **RON, octane sensitivity, HOV** (blue text).
- $\frac{d\eta_{glh}}{\eta_{glh}}$ (green circle) points to **Flame Speed** (green text).
- $\frac{d\eta_{pump}}{\eta_{pump}}$ (red circle) points to **HOV** (red text).
- $\frac{d\eta_{emiss}}{\eta_{emiss}}$ (purple circle) points to **PMI, $T_{c,90}$** (purple text).

How can we relate these terms to fuel properties?

Efficiency merit function approach



$$\begin{aligned} \text{Merit} = & \underbrace{\alpha \cdot f(\text{RON})}_{\text{RON}} + \underbrace{\beta \cdot f(K, S)}_{\text{Octane Sensitivity}} + \underbrace{\gamma \cdot f(\text{HOV})}_{\text{Heat of Vaporization}} \\ & + \underbrace{\varepsilon \cdot f(S_L)}_{\text{Flame Speed}} + \underbrace{\zeta \cdot f(\text{PMI})}_{\text{PM Emissions}} + \underbrace{\eta \cdot f(T_{c,90,conv})}_{\text{Catalyst Light-off Temp (cold start)}} \end{aligned}$$

Octane Index

Charge Cooling

Dilution Tolerance

Emissions Penalties

Efficiency merit function approach



$$\begin{aligned} \text{Merit} = & \frac{(RON_{mix} - 91)}{1.6} - K \frac{(S_{mix} - 8)}{1.6} \\ & + \frac{0.085[ON / kJ / kg_{mix}] \cdot ((HoV_{fuel} / (AFR_{mix} + 1)) - (415[kJ / kg] / (14.0[-] + 1)))}{1.6} \\ & + \frac{((HoV_{mix} / (AFR_{mix} + 1)) - (415[kJ / kg] / (14.0[-] + 1)))}{15.2} + \frac{(S_{Lmix} - 46[cm / s])}{5.4} \\ & - H(PMI_{mix} - 1.6)[0.7 + 0.5(PMI_{mix} - 1.4)] + 0.008^{\circ}C^{-1}(T_{c,90,conv} - T_{c,90,mix}) \end{aligned}$$

Technical report with details being published September 2017

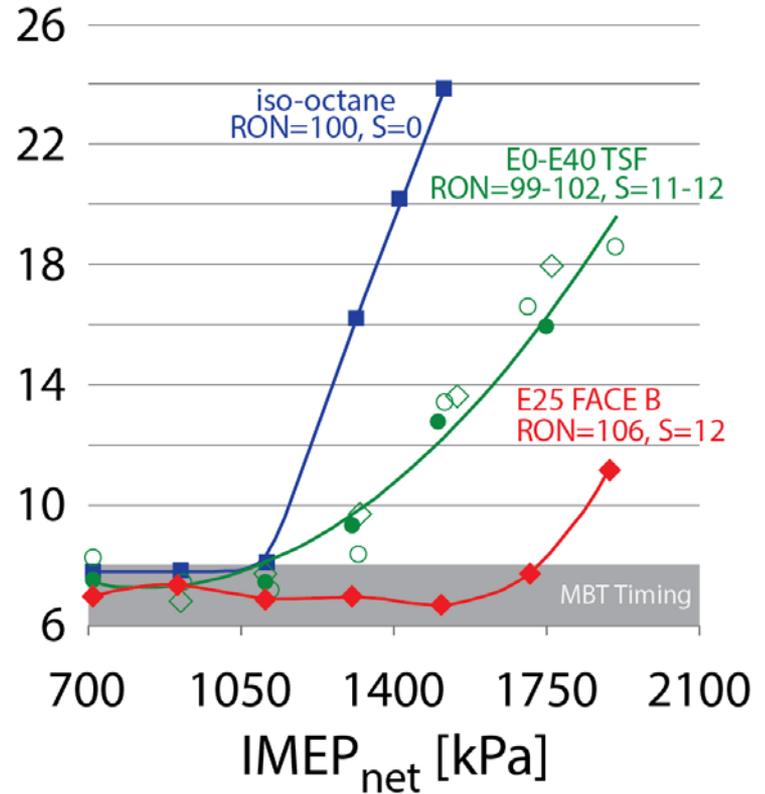
Decoupling S and HOV impacts



- With DI, increasing S from 0 to ~ 11 at RON=100 yields more than half the combustion phasing advance of RON 106, S~12 fuel
- All RON 100, S~11 fuels have similar knock-limited performance gains over ic8: no evident HoV benefit

Experimental details: Single cylinder version of GM Ecotec 2.0L, 9.2: CR; side-mounted DI or upstream fuel injection; load sweeps at an intake manifold temp of 50 °C; sweep intake manifold T for max load at 2 different CA50 phasing

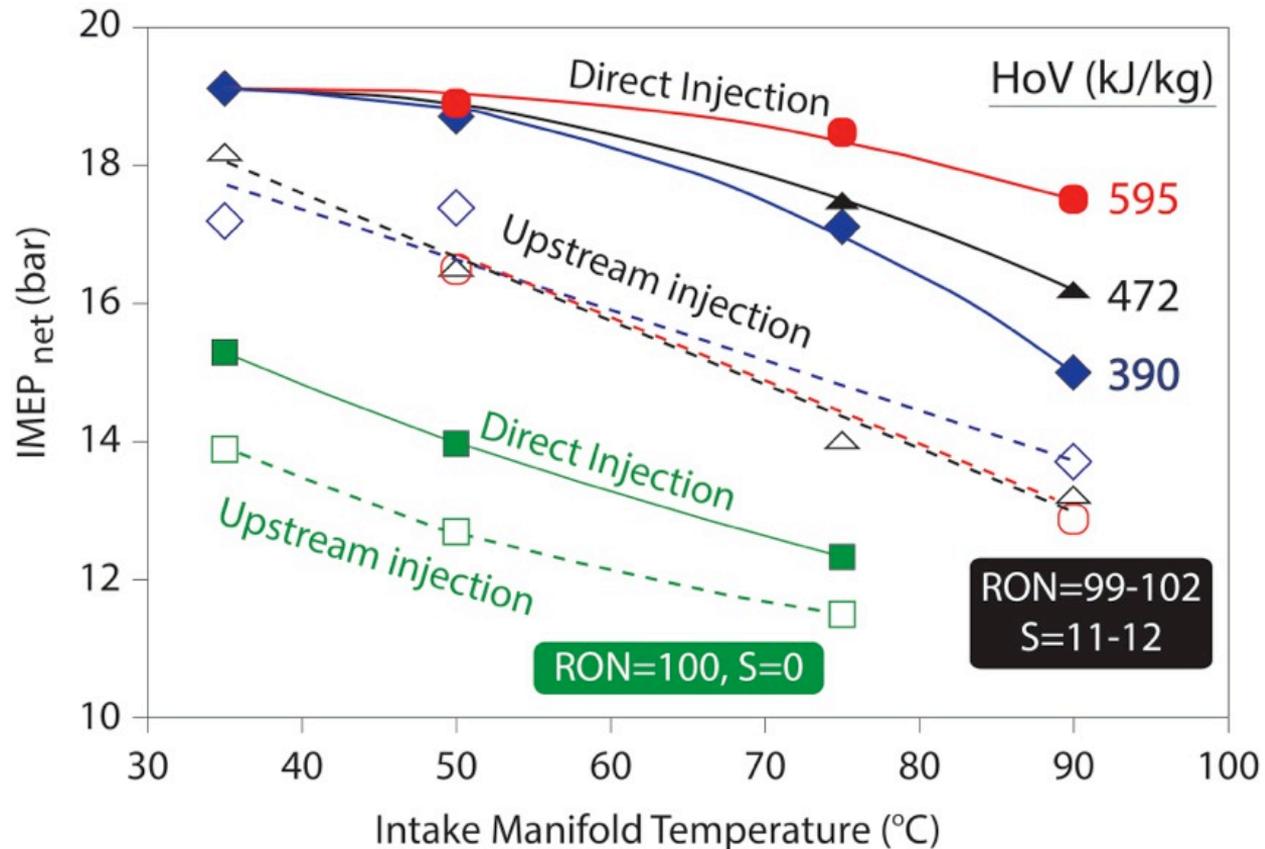
CA50 (atdc)



HoV can be important for DI at high intake T

Upstream injected (UI)
100 RON, $S \approx 11$ fuels
have higher peak IMEP at
constant CA50 than iso-
octane (RON 100, $S = 0$),
and HoV has little effect
(S is dominant)

- Direct injection (DI) of iso-octane has HoV benefit, but less than $S \approx 11$ effect
- DI of $S \approx 11$ fuels also has HoV benefit, which increases with manifold temp.



Question 2: What fuels should we make?



Approach:

Identify blendstock options that provide key properties



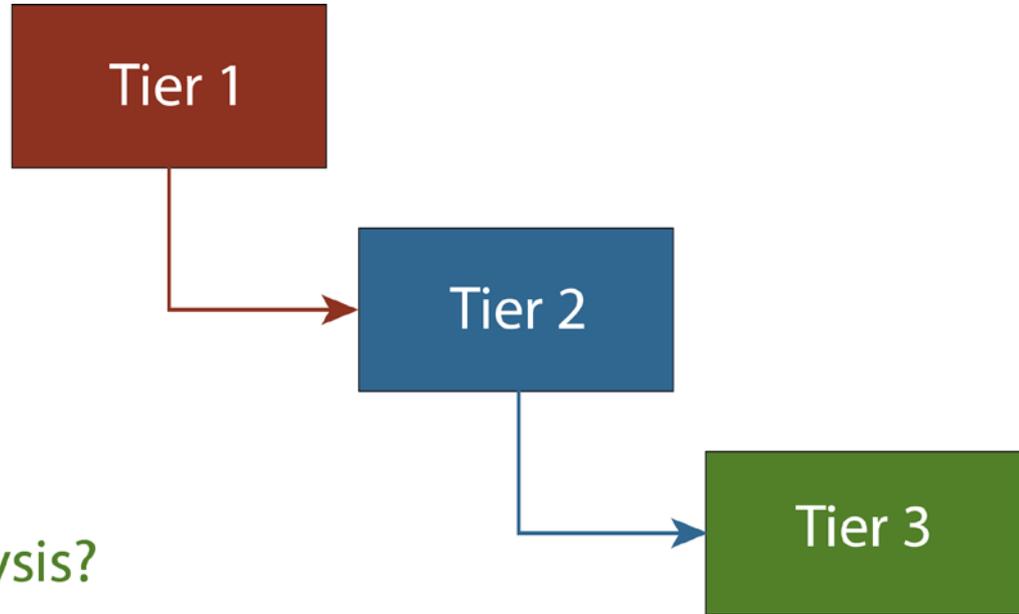
Tiered blendstock identification



Can it be a fuel blendstock?

Does it provide desired performance?

Does it merit focused experiments and analysis?



blendstocks:

> 400

~ 40

< 10

Tiered blendstock identification



Tier 1

> 470 blendstocks

14 chemical families

Identify broad range of potential hydrocarbon and oxygenated blendstocks

Utilize property information on blendstocks from literature or estimates to identify Tier 2 blendstocks

Hydrocarbons
Normal paraffins
Iso-paraffins
Cycloparaffins
Olefins
Aromatics
Multi-ring aromatics

Alcohols

Furans

Ethers

Carbonyls

Ketones

Aldehydes

Esters

Volatile fatty acid esters

Fatty esters

Carboxylic Acids

Present in commercial fuels

Not present in commercial fuels

A major goal of Co-Optima is to conduct a comprehensive and consistent survey of blendstock options:

What blendstocks are able to increase boosted SI performance?



Tiered blendstock identification



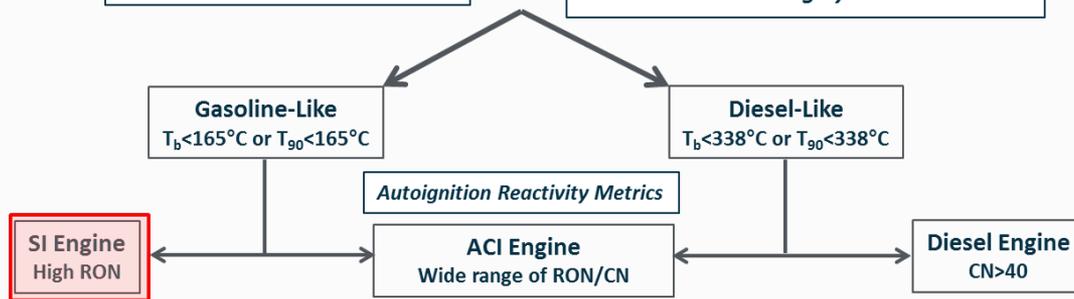
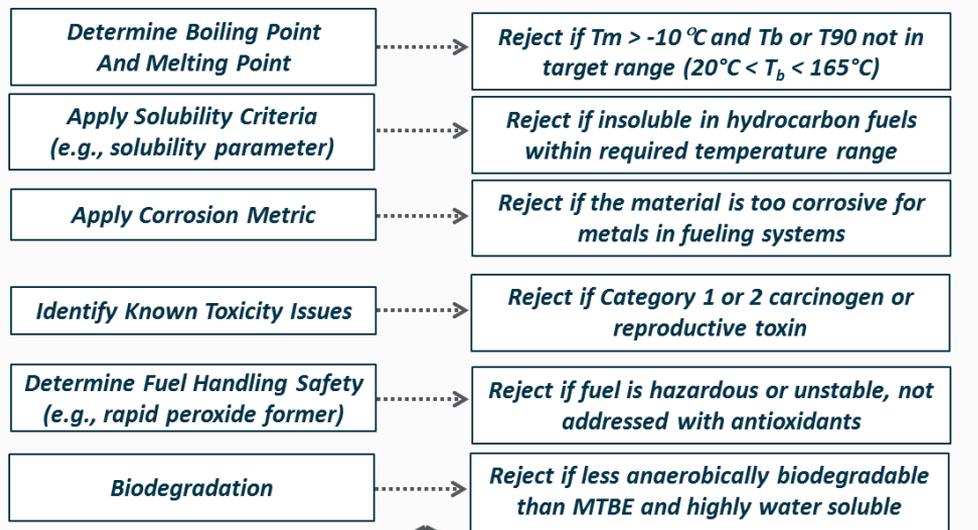
Tier 1

> 470 blendstocks

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Identify broad range of potential hydrocarbon and oxygenated blendstocks

Utilize property information on blendstocks from literature or estimates to identify Tier 2 blendstocks



Advanced SI Fuel Candidates



Tier 1 blendstock screening



Tier 1

> **470** blendstocks

14 chemical families

Identify broad range of potential hydrocarbon and oxygenated blendstocks

Utilize property information on blendstocks from literature or estimates to identify Tier 2 blendstocks

Tier 2

41 blendstocks

10 chemical families

Measure blendstock properties

Evaluate blendstock performance in BOBs at 10-30% blend levels

Remove candidates from list if improved data indicate they do not meet criteria

Add new candidates as our understanding improves of how fuel structure impacts key properties

Tier 3

Which blendstocks merit comprehensive, consistent, and rigorous study and analysis?



Boosted SI Tier 2 blendstocks



Alcohols (9)	
1	Methanol
2	Ethanol
3	1-Propanol
4	Isopropanol
5	1-Butanol
6	2-Butanol
7	Isobutanol
8	2-Methylbutan-1-ol
9	2-Pentanol

Ethers	
10	Anisole

Esters (13)	
11	Methyl acetate
12	Methyl butanoate
13	Methyl pentanoate
14	Methyl isobutanoate
15	Methyl-2-methylbutanoate

Esters (13)	
16	Ethyl acetate
17	Ethyl butanoate
18	Ethyl isobutanoate
19	Isopropyl acetate
20	Butyl acetate
21	2-Methylpropyl acetate
22	3-Methylpropyl acetate
23	mixed esters

Ketones (9)	
24	2-Butanone
25	2-Pentanone
26	3-Pentanone
27	Cyclopentanone
28	3-Hexanone
29	4-Methyl-2-Pentanone
30	2,4-Dimethyl-3-Pentanone
31	3-Methyl-2-butanone
32	Ketone mixture

Furans	
33	2,5-Dimethylfuran/2-methylfuran

Branched alkanes	
34	2,2,3-Trimethylbutane

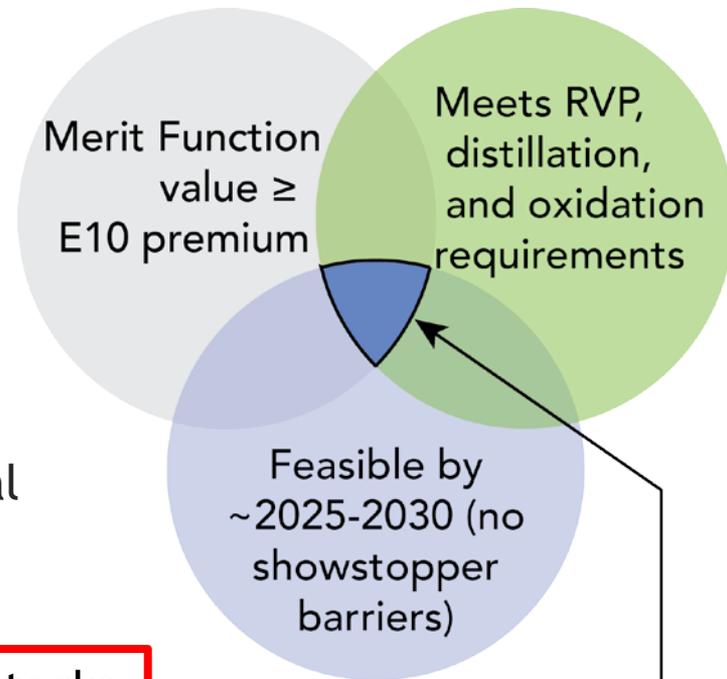
Alkenes	
35	Diisobutylene

Multicomponent mixtures (6)	
36	Methanol-to-gasoline
37	Ethanol-to-gasoline
38	Bioreformate via multistage pyrolysis
39	Bioreformate via catalytic conversion of sugar
40	Mixed aromatics via catalytic fast pyrolysis
41	Aromatics and olefins via pyrolysis-derived sugars

Tier 2 to Tier 3 transition criteria



1. Achieve merit function score \geq E10 premium when blended in petroleum BOB*
2. Meet current critical fuel specs (RVP, distillation, oxidative stability, etc.) when blended in petroleum BOB*
3. No “showstopper” barriers
 - Candidates must have viable path to potential market introduction by ~2025 - 2030

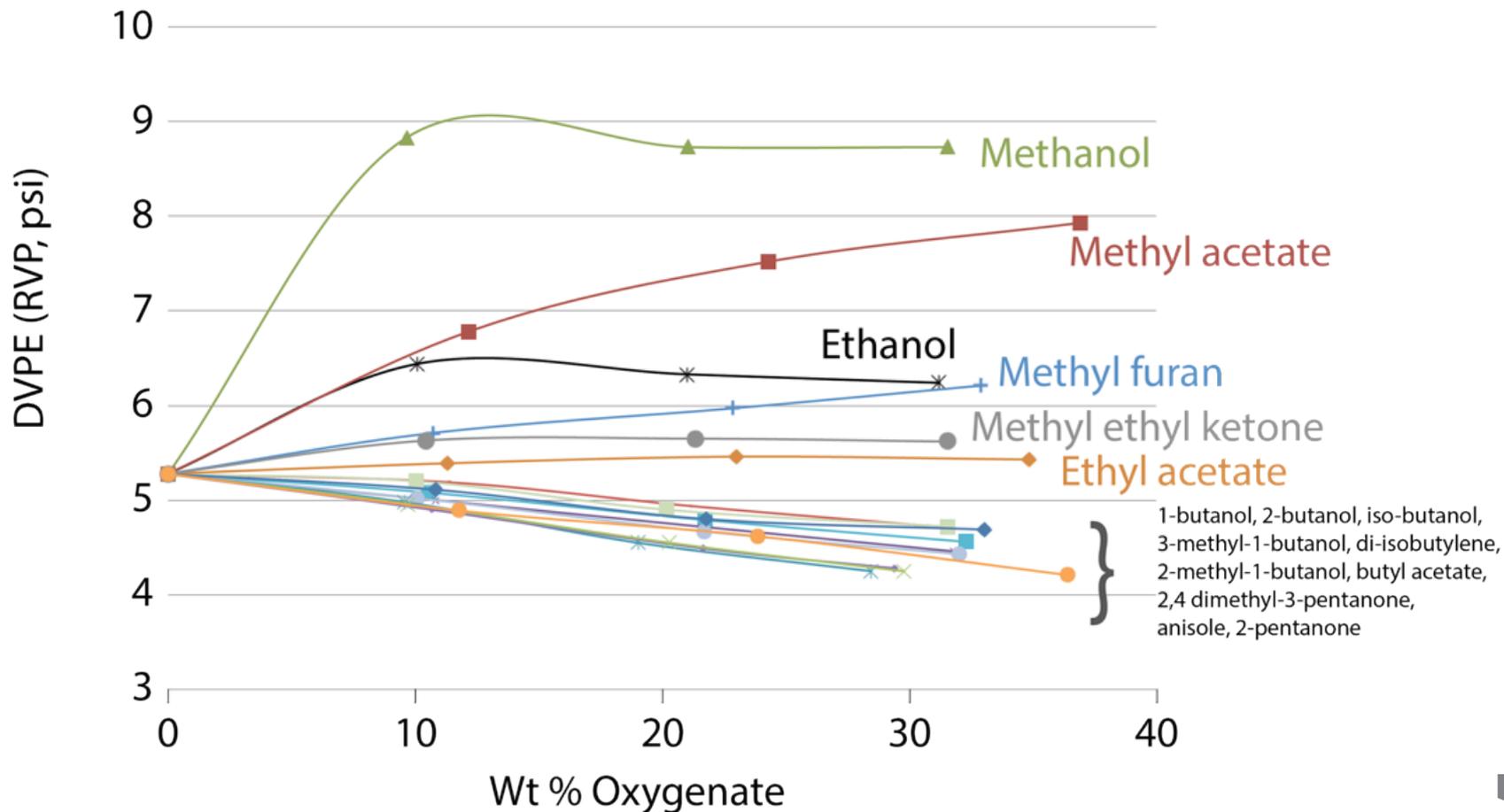


Tier 3 blendstocks

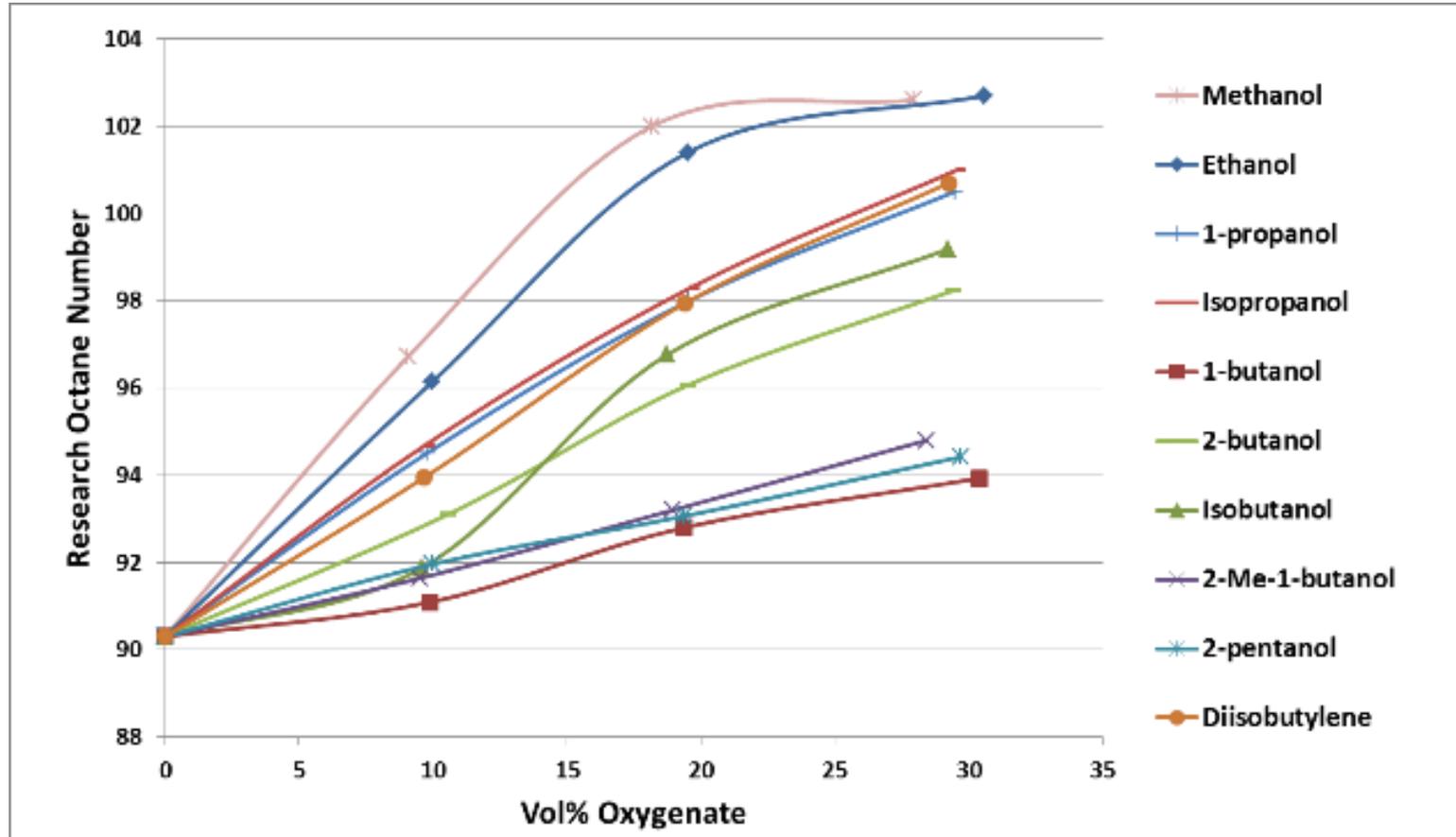
Tier 2->3 transition allows focused effort on blendstocks with greatest potential to meet Co-Optima goals

* BOB = blendstock for oxygenate blending; evaluated at blend levels of 10, 20, and 30% by volume

Example blendstock data: RVP



Determining blending behavior



High-potential blendstocks identified

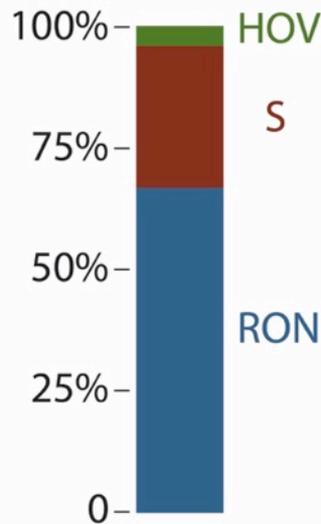


Properties provided by chemical families:

RON S HOV

Alcohols	✓	✓	✓
Furans	✓	✓	
Alkenes	✓	✓	
Aromatics	✓	✓	
Ketones	✓	✓	
Cycloalkanes	✓	✓	
Alkanes	✓		
Ethers	✓		
Esters	✓		

Average contribution to merit function for highest scoring blendstocks



RON = Research octane number ; S = Sensitivity ($S = RON - MON$) ; HOV = heat of vaporization

Question 3: What will work in the real world?



Approach:

Conduct comparative, systems-level analyses of economic, environmental, state of technology, and market factors

Assess likelihood of commercial scale impact by 2025-2030



Analysis Metrics



Technology Readiness

State of technology:
Fuel production

State of technology:
Vehicle use

Conversion technology
readiness level

Feedstock sensitivity

Process robustness

Feedstock quality

of viable pathways



Environmental

Carbon efficiency

Target yield

Life cycle greenhouse
gas emissions

Life cycle water

Life cycle fossil
energy use



Economics

Target cost

Needed cost reduction

Co-product economics

Feedstock cost

Alternative high-value
use



Market

Uncertainty

Regulatory requirements

Geographic factors

Political factors

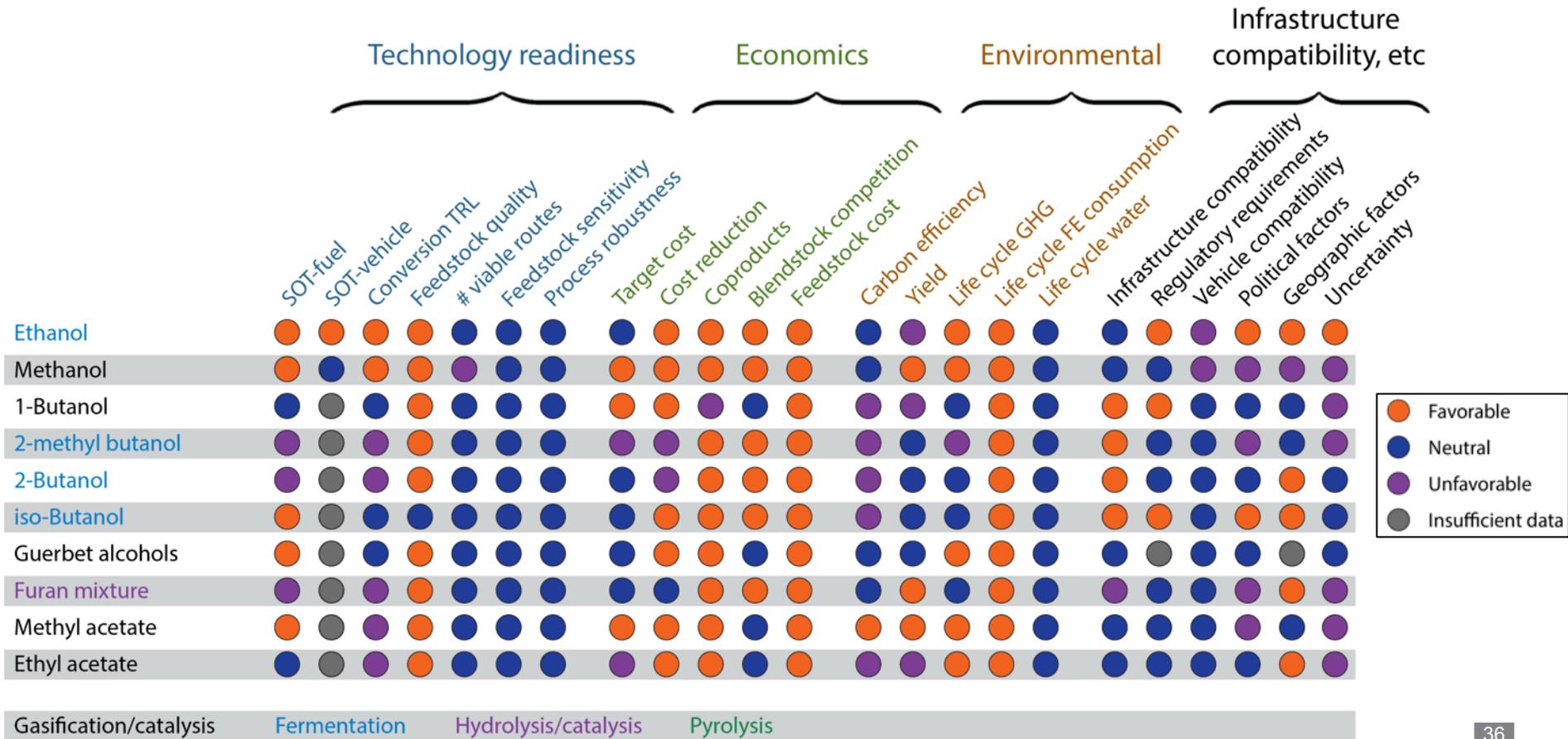
Vehicle compatibility

Infrastructure
compatibility

Assessed only for blendstocks
produced from biomass

Assessed for both fossil
and renewable blendstocks

Screening assessment results



High-potential blendstocks identified

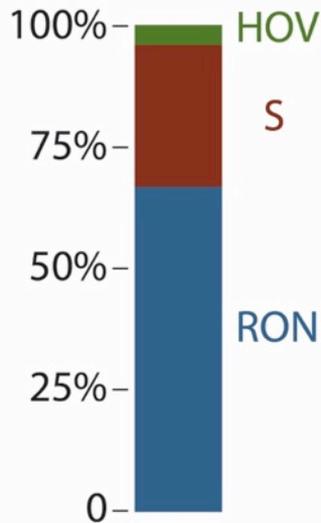


Properties provided by chemical families:

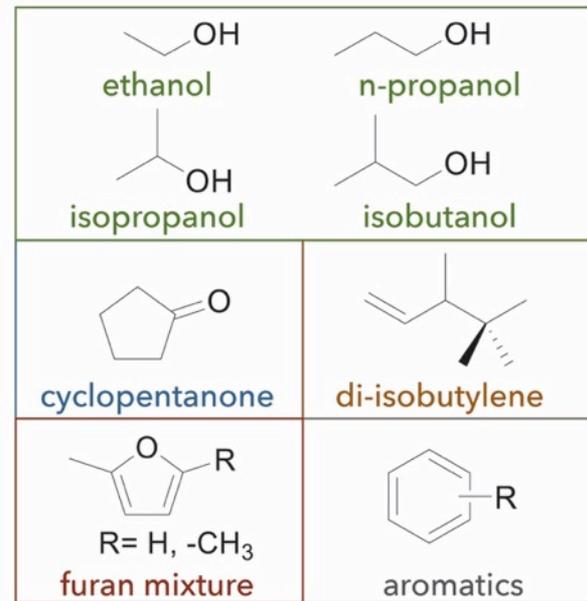
RON S HOV

Alcohols	✓	✓	✓
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Alkenes	✓	✓	
Aromatics	✓	✓	
Ketones	✓	✓	
Cycloalkanes	✓	✓	
Alkanes	✓		
Ethers	✓		
Esters	✓		

Average contribution to merit function for highest scoring blendstocks



Eight representative blendstocks selected for more detailed evaluation



RON = Research octane number ; S = Sensitivity (S = RON – MON) ; HOV = heat of vaporization

Current boosted SI blendstock efforts

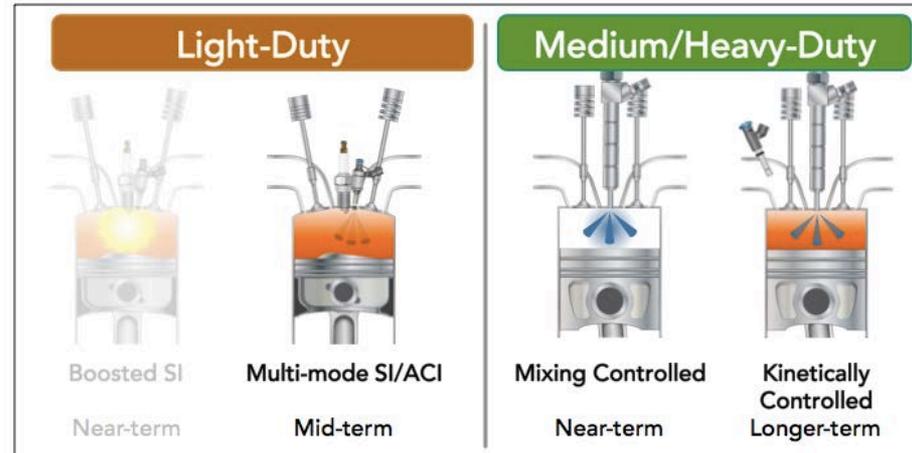


- Refine property measurements, improve blend models, and conduct more detailed compatibility studies
- Conduct engine tests to confirm performance and assess potential to meet FE targets
- Carry out emissions control experiments to assess impacts on efficiency and durability
- Conduct detailed life cycle, techno-economic analyses, and refinery integration studies

Next steps



- Refine merit function and establish technical basis for advanced gasoline fuel specification for boosted SI by end of FY18
- Conduct more rigorous assessments of Tier 3 candidates
- Assess candidates for potential follow-on scale-up studies
- Expand LD efforts – multi-mode SI-ACI
- Expand MD/HD efforts
- Continue strong engagement with stakeholders to help focus R&D on options that provide “wins” for broad range of stakeholders



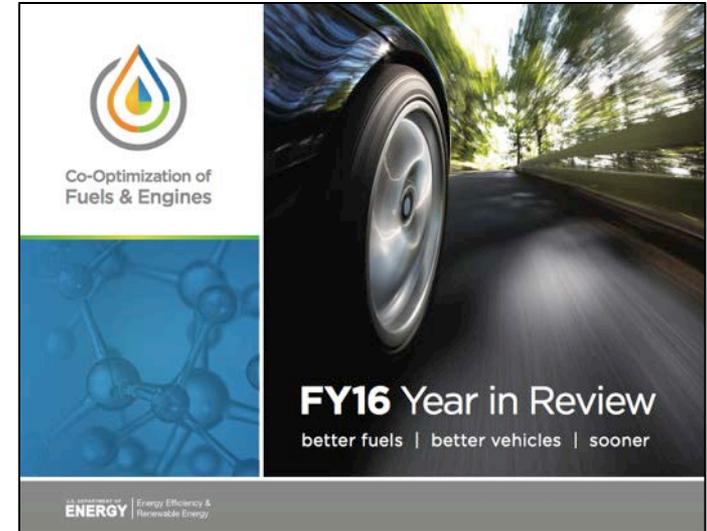


Co-Optima Website



<https://energy.gov/eere/bioenergy/co-optimization-fuels-engines>

FY 2016 Year in Review Highlights



<https://www.nrel.gov/docs/fy17osti/67595.pdf>

2017 VTO Peer Review Presentations



Detailed overview available from FY17 VTO AMR presentations

<https://www.annualmeritreview.energy.gov>

2017 Bioenergy Peer Review Presentations



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2017 Project Peer Review—Co-Optimization of Fuels and Engines

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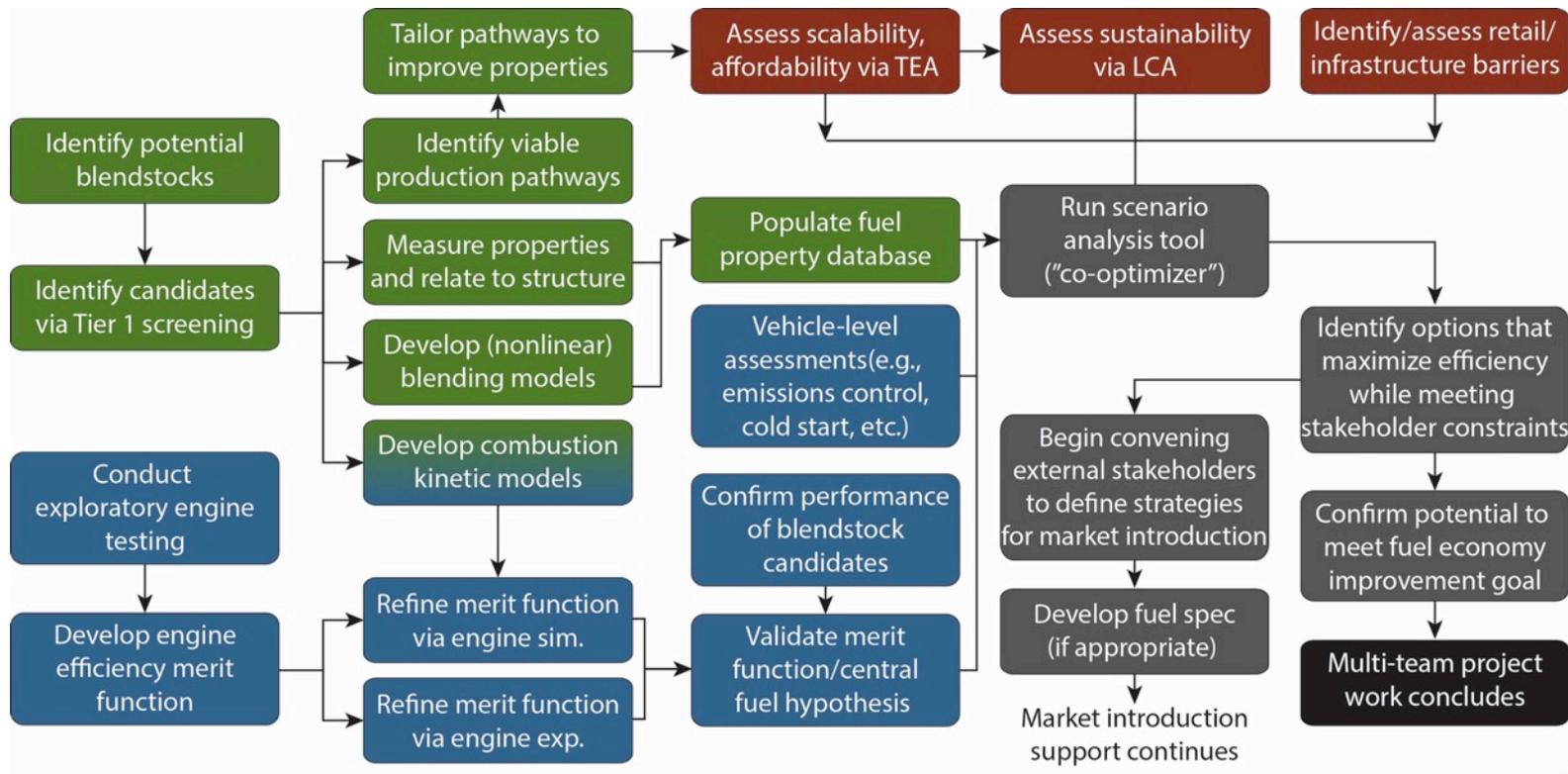
The Bioenergy Technologies Office hosted its 2017 Project Peer Review on March 6-9, 2017, in Denver, Colorado. The presentations from the Co-Optimization of Fuels and Engines sessions are available to view and download below. For detailed session descriptions and presentation titles, view the [2017 Project Peer Review Program Booklet](#).

- [Co-Optima Overview](#)
- [High-Performance Fuels](#)
- [Analysis of Sustainability, Supply, Economics, Risk and Trade \(ASSERT\)](#)
- [Market Transformation](#)



Thank You!

Technical Approach



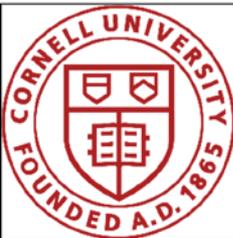
Map properties to efficiency
"What fuels to engines want?"

Expand blendstock options
"What fuels should we make?"

Identify barriers to use
"What will work in real world?"

Identifying options
"How do we co-optimize?"

University Partners



Cornell / UCSD

Identify differences in combustion characteristics of diesel/biofuel blends vs petroleum-based fuels



LSU / TAMU / U Conn.

Develop method to characterize alternative fuel candidates and associated models and metrics for predicted engine performance



Univ. Michigan

Develop engine combustion model to simulate key parameters while reducing computational expense 80%



MIT / Univ. Central Florida

Develop detailed kinetic models for several biofuels using an advanced computational approach



U. Mich. - Dearborn/Oakland U.

Use a miniature ignition screening RCM to study ignition properties/combustion characteristics of alternative fuels.



Yale

Measure sooting tendencies of various biofuels and develop emission indices relevant to real engines



Univ. Alabama

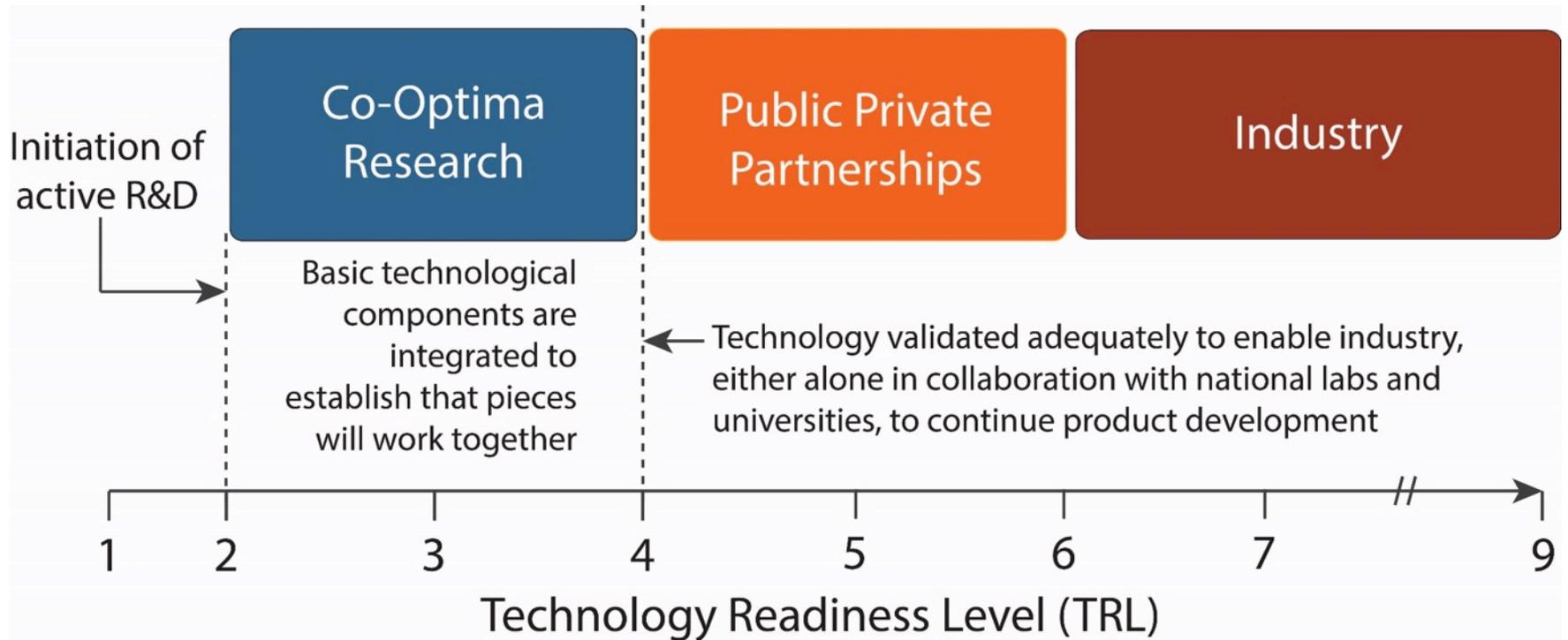
Examine combustion properties of biofuels and blends using advanced diagnostics under realistic ACI engine conditions.



Univ. Central FL/Penn State

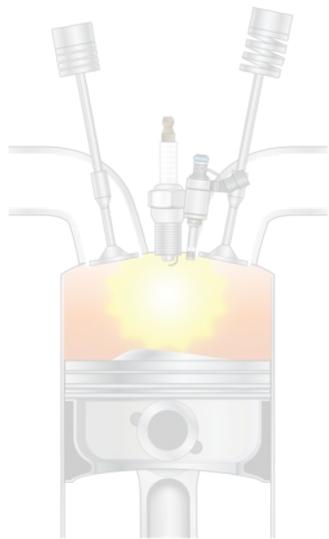
Generate fuel characterization data related to fuel spray atomization, flame topology, etc, and compatibility for prioritized fuels

Integration With Industry



See https://en.wikipedia.org/wiki/Technology_readiness_level

Light-Duty



Boosted SI

Near-term



Multi-mode SI/ACI

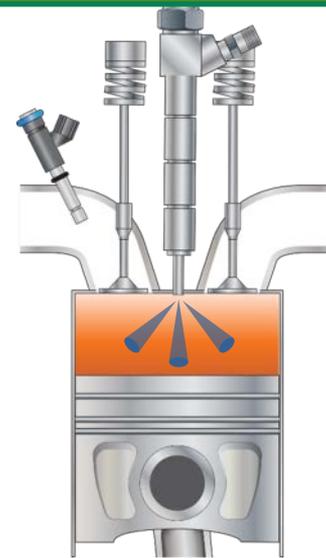
Mid-term

Medium/Heavy-Duty



Mixing Controlled

Near-term



**Kinetically
Controlled**
Longer-term

Fuel property database*



Tier 1

> 470 blendstocks

14 chemical families

Identify broad range of potential hydrocarbon and oxygenated blendstocks

Utilize property information on blendstocks from literature or estimates to identify Tier 2 blendstocks

The screenshot displays the 'Found Pure Compound' interface for 1,4-Pentanediol. At the top, it says 'Found Pure Compound' in red, with a 'Correct or Update this record' button. Below this, the IUPAC name '1,4-Pentanediol' is entered in a search box. The interface shows various fields for molecular weight (104.15), molecular formula (C5H12O2), CAS# (626-95-9), and functional group. A chemical structure of 1,4-pentanediol is shown on the right. Below the structure, there is a 'SEARCH PROPERTIES' section with a search box and a 'Properties' section with a grid of input fields for various physical and chemical properties such as Melting Point, Boiling Point, Vapor Pressure, and Density.

* Publicly accessible: <https://fuelsdb.nrel.gov/fmi/webd#FuelEngineCoOptimization>



Tier 1 blendstock screening



Tier 1

> 470 blendstocks

14 chemical families

Identify broad range of potential hydrocarbon and oxygenated blendstocks

Utilize property information on blendstocks from literature or estimates to identify Tier 2 blendstocks

Hydrocarbons
Normal paraffins
Iso-paraffins
Cycloparaffins
Olefins
Multi-ring aromatics
Alcohols
Furans
Ethers
Carbonyls
Ketones
Aldehydes
Esters
Volatile fatty acid esters
Fatty esters
Carboxylic Acids

YES

Normal paraffins
Iso-paraffins
Cycloparaffins
Olefins
Alcohols

YES FOR SOME

Aromatics
Ketones
Volatile fatty acid esters
Furans
Ethers

NO

Multi-ring aromatics
Aldehydes
Fatty esters
Carboxylic acids

